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# STUDIES ON COPPER METABOLISM. XI. COPPER AND IRON METABOLISM IN THE NEPHROTIC SYNDROME<sup>1</sup>

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The metabolism of copper and iron in the nephrotic syndrome has received little attention. In view of the fact that copper in the plasma is bound almost exclusively to an  $\alpha_2$ -globulin (ceruloplasmin) with a molecular weight of approximately 151,000 (1, 2) while iron is bound to a  $\beta_1$ -globulin (transferrin) with a molecular weight of about 90,000 (3), it would seem not unlikely that these two metal-binding proteins might be excreted in large quantities in the urine of patients with pronounced proteinuria. If this loss were great and extended over a prolonged period, it is conceivable that depletion of the body stores of these two elements might occur or the capacity of the body to synthesize these two proteins might be exceeded, with the consequence that the plasma level of ceruloplasmin and transferrin would be reduced.

In the course of investigations in this laboratory concerning the anemia associated with infection (4), a patient was studied who, while afflicted with chronic osteomyelitis, developed the nephrotic syndrome. In association with the severe hypoproteinemia and proteinuria, the plasma copper level decreased from 246  $\mu\text{g}$ . per 100 ml., the high levels usually found with infection, to 40  $\mu\text{g}$ . per 100 ml. in spite of the continued presence of osteomyelitis. More recently (5) we observed hypocupremia in two of three children with the nephrotic syndrome. Munch-Petersen (6) has reported that the urinary excretion of copper is increased in patients with proteinuria and that the amount of copper excreted is greatest in those with the highest concentration of protein in the urine. Although he did not record the plasma copper values he stated that there was no correlation between the plasma copper concentration and the amount of copper in the urine.

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Hypoferremia was noted in association with the nephrotic syndrome in one patient by Laurell (7) and was observed by us in two of three children with this condition (5). Laurell found an extremely low serum iron-binding capacity in his patient. Slater and Kunkel (8), using paper electrophoresis, observed a distinct  $\beta$ -globulin spot in the urine of patients with nephrosis, sometimes without a visible counterpart in the serum. They suggested that this  $\beta$ -globulin might be serum iron-binding  $\beta_1$ -globulin.

It is the purpose of this paper to present studies on the plasma copper and iron levels and the excretion of these two elements in the urine of patients with the nephrotic syndrome.

## METHODS

The hematologic methods used in this study have been described elsewhere (9). The methods employed have been described in other communications; for the measurement of red blood cell and plasma copper (10); for the direct-reacting copper fraction of plasma (11); for plasma iron (12); and for total iron-binding capacity of the plasma (13). Total serum protein, albumin and globulin were determined by the biuret method (14) with the modification of Weichselbaum (15). The electrophoretic analyses were performed as described previously (16).

The protein concentration in the urine was determined spectrophotometrically by a modification of the biuret method of Hiller, Greif, and Beckman (17). Sufficient urine to contain 5 to 20 mg. of protein was pipetted into a medium-sized pyrex test tube. An equal volume of Tsuchiya's reagent (18) was added to precipitate the proteins. The tubes were centrifuged at 2000 to 2500 r.p.m. for 10 minutes, after which the supernate was decanted off and the tubes drained by inverting on a paper towel. The pellet of protein was dissolved by mixing with 2 ml. of a 4 per cent solution of sodium hydroxide. Three ml. of distilled water were then added, followed by 5 ml. of the dilute biuret reagent described by Weichselbaum (15). The mixture was allowed to stand 20 to 30 minutes for color development, after which the optical density of the sample, as well as that of a blank prepared in the same manner but without urine, was read by

the use of an Evelyn colorimeter with a 565  $m\mu$  filter. If the urine was highly colored, a urine blank was read with 5 ml. of a 0.2 per cent solution of sodium hydroxide added in place of the biuret reagent. The protein concentration was obtained from a curve relating the optical density to protein content as obtained by the use of various dilutions of serum, the Kjeldahl method being the standard of reference (19).

For the determination of urinary copper, 20 ml. aliquots of urine were transferred to 100 ml. Kjeldahl flasks. To each flask were added 1.5 ml. concentrated sulfuric acid and 10 ml. of concentrated nitric acid. The mixture was digested until charred. It was then cooled, and one ml. of concentrated perchloric acid and 5 ml. of concentrated nitric acid were added to each flask, the digestion being continued until the solution became water-clear. The contents were partially cooled, transferred to 10 ml. volumetric flasks and made up to volume with redistilled water. A blank was prepared in a similar manner. To one ml. aliquots were added 1.2 ml. of a saturated solution of sodium citrate, 0.2 ml. of a one per cent solution of gum arabic and 0.8 ml. of concentrated ammonium hydroxide. The density ( $D_1$ ) of the solution was read in a Beckman spectrophotometer at a wave length of 440  $m\mu$ . Two-tenths ml. of a 0.1 per cent solution of sodium diethyldithiocarbamate was added and the density ( $D_2$ ) determined. The density ( $D_3$ ) of the blank was then read. The concentration of copper was calculated from the following formula:

$$\text{Cu in } \mu\text{g./100 ml. urine} = K [D_2 - (3.2/3.4 D_1 + D_3)],$$

where K is the constant derived from the standard curve prepared from analytical grade metallic copper.

For the determination of urinary iron one ml. of the above described diluted digest was pipetted into a small test tube and 2 ml. of saturated aqueous sodium acetate solution added to make the solution basic (red) to congo red paper. One-tenth ml. of thioglycolic acid and 0.4 ml. of a 0.1 per cent solution of O-phenanthroline were added. A blank was prepared in a similar manner. The mixture was allowed to stand one hour for color development after which time the optical density ( $D_2$ ) was read in the Beckman spectrophotometer at a wave length of 510  $m\mu$ . The density ( $D_3$ ) of the blank solution was determined and the concentration of iron was calculated from the following formula:

$$\text{Fe in } \mu\text{g./100 ml. urine} = K (D_2 - D_3),^2$$

where K is the constant derived from a standard curve prepared from pure iron wire.

Dialysis of urine copper and iron was carried out in cellophane tubing which had been previously washed with dilute acid and thoroughly rinsed with distilled water. This was done for 18 hours at 2° C., with occasional agitation. The dialysate was changed twice during the 18 hours. After completion of dialysis the volume of the

sample in the tubing was measured and the copper and iron content of the dialyzed urine were determined as well as that of the urine prior to dialysis.

All of the patients with the nephrotic syndrome studied manifested the classical features of this condition—namely, intractable edema, massive proteinuria, lipemia, and hypoalbuminemia (Table II). The nephrosis in the adults presumably was secondary to chronic glomerulonephritis although in not all of the patients was such a history suggestive of that disease obtainable. In none was the nephrosis associated with intercapillary glomerulo-sclerosis, amyloidosis, syphilis, renal vein thrombosis, or other known rare causes of this syndrome. In all of the children studied the syndrome could be classified as the so-called "pure" or "lipoid nephrosis."

One of the adult patients (E. C.) was somewhat unusual and difficult to classify since, in addition to most of the features of the nephrotic syndrome, there was a moderately severe anemia, elevated blood urea nitrogen, failure to concentrate the urine and other evidence of impaired renal function, and a urinary tract infection. In certain of the other patients complications were present, as indicated in Table II.

Six of the patients (C. E., R. K., G. H., W. B., D. G., and E. C.) were studied on a metabolic ward for at least several weeks. The recorded values represent the means of many determinations. In the case of the children, on the other hand, a single blood specimen was studied.<sup>3</sup>

Creatinine values were determined on all specimens of urine.

## RESULTS

### *Plasma copper, plasma iron, plasma protein, and total iron-binding capacity of the plasma in normal human subjects*

Plasma copper values in a total of 218 normal individuals have been determined in this laboratory and have been reported previously (4, 20, 21). The values for 10 additional male subjects are given in Table I. The mean value ( $\pm$  S.D.) for the entire group of 228 adult subjects is  $116 \pm 14$   $\mu\text{g. per 100 ml.}$  In only 10 subjects (4.4 per cent) has a value of less than 90  $\mu\text{g. per 100 ml.}$  been obtained. In two of the ten, the values were less than 80  $\mu\text{g. per 100 ml.}$ , both being 68. Unpublished observations of Dr. Phillip Sturgeon, who has used our method, showed the mean values in 17 children, 3 to 10 years of age, to be  $131 \pm 27$   $\mu\text{g. per 100 ml.}$  (range 99 to 188).

Plasma iron values in a total of 169 normal individuals have been reported previously from this

<sup>2</sup> The  $D_1$  reading (urine blank) was negligible at 510  $m\mu$  and, therefore, has been omitted from the formula.

<sup>3</sup> We are indebted to Dr. Phillip Sturgeon, Children's Hospital Society, Los Angeles, Cal., for sending us the case histories and specimens from seven of the children.

TABLE I  
*Proteinuria, cupriuria, and siderinuria in ten normal male subjects*

Subject	Plasma					Urine		
	Copper	Iron	Total protein	Albumin	Globulin	Copper	Iron	Protein
	$\mu\text{g./100 ml.}$	$\mu\text{g./100 ml.}$	$\text{Gm./100 ml.}$	$\text{Gm./100 ml.}$	$\text{Gm./100 ml.}$	$\mu\text{g./24 hr.}$	$\mu\text{g./24 hr.}$	$\text{Gm./24 hr.}$
1	104	95	6.2	3.9	2.3	0	40	0.04
2	101	86	7.0	4.1	2.9	0	36	0.05
3	113	119	6.8	4.1	2.7	11	54	0.07
4	107	83	6.5	3.8	2.7	10	32	0.08
5	137	118	6.6	4.0	2.6	8	54	0.04
6	113	192	6.4	4.3	2.1	0	62	0.03
7	104	171	6.6	4.2	2.4	12	57	0.02
8	92	69	6.5	4.0	2.5	26	43	0.04
9	96	100	7.2	4.3	2.9	15	39	0.02
10	99	135	6.1	4.1	2.0	11	64	0.06
Mean	107	117	6.6	4.1	2.5	9	48	0.05

laboratory (4, 13, 21, 22). The values for 10 additional normal subjects are presented in Table I. The mean value ( $\pm$  S.D.) for the entire group of 179 individuals is  $110 \pm 31 \mu\text{g.}$  per 100 ml. Values of less than  $70 \mu\text{g.}$  per 100 ml. have been observed in 15 subjects (8 per cent). In four females and one male, values between 39 and  $50 \mu\text{g.}$  have been obtained. In 17 children, 3 to 10 years of age, Dr. Phillip Sturgeon has found the mean serum iron to be  $86 \pm 33 \mu\text{g.}$  per 100 ml. (range 27 to 153) (Unpublished observations).

Plasma total protein, albumin, and globulin values for 10 normal male subjects are recorded in Table I. The mean values were 6.6, 4.1, and 2.5 Gm. per 100 ml., respectively.

The total iron-binding capacity of the serum as measured in this laboratory in 30 normal adults was  $359 \pm 30.8 \mu\text{g.}$  per 100 ml. (13). Dr. Phillip Sturgeon's value for 13 children 3 to 10 years of age is  $404 \pm 118 \mu\text{g.}$  per 100 ml. No significant difference between the total iron-binding capacity of the serum and of the plasma has been observed.

TABLE II  
*Nephrotic syndrome*

Patient*	Age in years	Sex	V.P.R.C.†	BUN†	Serum cholesterol	Plasma protein	Plasma albumin	Plasma globulin	Plasma copper	Plasma iron	T.I.B.C.†
			$\text{ml./100 ml.}$	$\text{mg./100 ml.}$	$\text{mg./100 ml.}$	$\text{Gm./100 ml.}$	$\text{Gm./100 ml.}$	$\text{Gm./100 ml.}$	$\mu\text{g./100 ml.}$	$\mu\text{g./100 ml.}$	$\mu\text{g./100 ml.}$
C. E. <sup>1</sup>	45	F	33	8	680	4.2	0.9	3.3	63	33	83
R. K.	24	M	42	6	700	4.2	0.9	3.3	58	64	64
G. H. <sup>2</sup>	26	M	30	8	900	3.5	0.7	2.8	94	95	
W. B. <sup>3</sup>	42	M	30	46	403	5.3	1.8	3.5	89	96	
D. G. <sup>4</sup>	73	M	43	70	590	4.3	1.3	3.0	49	42	92
E. C. <sup>1,5</sup>	58	F	26	82	350	4.6	2.3	2.3	20	117	
M. W.	10	M	46	54	700	4.3	2.2	2.1	60	30	
K. J.	4	M	47	40	730	4.8	1.9	2.9	96	58	
A. H. <sup>1</sup>	3	M	40	32	510	5.1	3.0	2.1	84	97	
M. G.	7	F	33	27	955	3.2	0.9	2.3	68	23	23
A. G.	7	M	38	16	514	3.8	1.1	2.7	56	26	106
R. M.	3	M	35	10	955	4.1	1.4	2.7	59	26	
M. F.	5	M	38	22	535	3.2	1.2	2.0	50	24	
C. W.	5	M	35	31	630	3.6	1.3	2.3	63	16	91
C. B.	5	M	26	26	364	3.7	1.2	2.5	47	21	
D. B.	5	M	40	15	807	4.1	1.8	2.3	69	33	88
Mean			36	31	645	4.1	1.5	2.6	64	50	78
S.D.			6.6	7.1	195	0.62	0.62	0.47	19.6	33.3	
Range			26-47	6-82	350-995	3.2-5.3	0.7-3.0	2.0-3.5	20-96	16-117	23-106

\* <sup>1</sup> Associated urinary tract infection; <sup>2</sup> associated bronchiectasis; <sup>3</sup> pneumonia; <sup>4</sup> gout; <sup>5</sup> uremia complicating the nephrotic syndrome.

† V.P.R.C., volume of packed red cells; BUN, blood urea nitrogen; T.I.B.C., total iron binding capacity of the plasma.

### Urinary copper, iron and protein in normal human subjects

The values for urinary copper, iron and protein in 10 normal male adults are given in Table I together with the plasma values of copper, iron, and protein in these individuals.

The mean excretion of copper per 24 hours was 9  $\mu\text{g.}$ , with a range from 0 to 26. The mean excretion of iron per 24 hours was 48  $\mu\text{g.}$ , with a range from 32 to 64  $\mu\text{g.}$  The excretion of protein was negligible, with a mean value of 0.05 Gm. per 24 hours and a range of 0.02 to 0.08.

### Plasma copper, plasma iron and the total iron-binding capacity of the plasma in the nephrotic syndrome.

The values for plasma copper<sup>a</sup> and iron for sixteen patients with the nephrotic syndrome are presented in Table II. The mean value ( $\pm$  S. D.) for the group was  $64 \pm 20$   $\mu\text{g.}$  per 100 ml. The value observed for plasma copper was more than two standard deviations below the normal mean in 13 of the 16 patients. In 10 of the 16 patients the plasma copper levels were lower than the lowest value observed in a normal subject.

The mean value ( $\pm$  S.D.) for the plasma iron

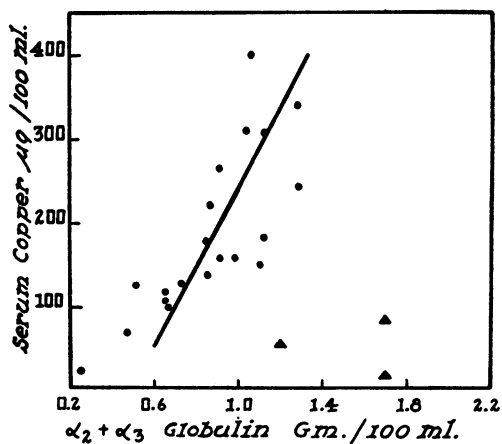


FIG. 1. THE CORRELATION BETWEEN THE PLASMA COPPER LEVEL AND THE  $\alpha_2 + \alpha_3$  GLOBULIN FRACTION OF SERUM IN NORMAL HUMAN SUBJECTS AND SUBJECTS WITH HYPERCUPREMIA (PREGNANCY, INFECTIONS AND LEUKEMIA) (Dots)

In the three patients (C. E., D. G., and E. C.) with the nephrotic syndrome ( $\blacktriangle$ ) the serum copper levels were low but the globulin level was high. The diagonal line was plotted by the method of least squares.

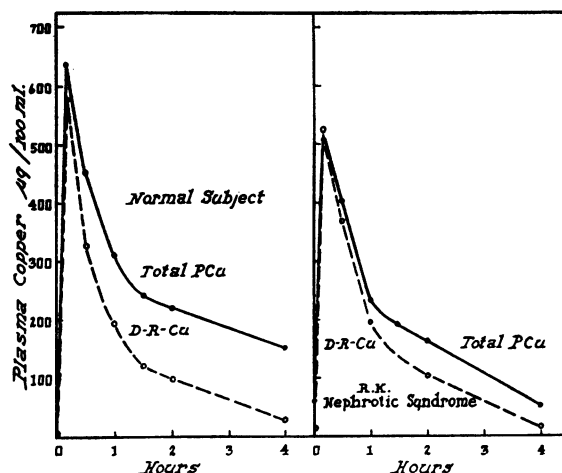


FIG. 2. THE REMOVAL OF COPPER FROM THE PLASMA FOLLOWING THE INTRAVENOUS ADMINISTRATION OF 100 MG. OF "CUPRALENE" TO A NORMAL SUBJECT AND TO A PATIENT WITH THE NEPHROTIC SYNDROME

The rate of removal of copper during the first 1.5 hours was the same in each individual. In both the normal subject and the patient most of the copper injected was in the direct-reacting (D-R-Cu) fraction.

level for the entire group was  $50 \pm 33$   $\mu\text{g.}$  per 100 ml. The mean value in the adults was 74  $\mu\text{g.}$  per 100 ml. and in the children it was 35  $\mu\text{g.}$  per 100 ml.

Due to the lipemic nature of the plasma, the total iron-binding capacity could be determined reliably in only seven of the patients. In all of these there was a marked reduction but the mean per cent saturation of the protein with iron was 53 per cent (26 to 100) as compared with a mean of 35 per cent (26 to 49) for normal subjects (13).

Red blood cell copper determinations were performed in five patients. In two of these (C. E. and W. B.) the values were normal, 94 and 87  $\mu\text{g.}$  per 100 ml. of packed RBC, respectively. In two (R. K. and D. G.) the values of 61 and 71  $\mu\text{g.}$  per 100 ml. were more than two standard deviations below the normal male mean of 110  $\mu\text{g.}$  per 100 ml. (21). In the fifth patient (E. C.) the extremely low value of 40  $\mu\text{g.}$  was obtained. It may be significant that this patient had the lowest plasma copper level of all of the patients studied.

In a previous communication (16) it was noted that in normal subjects and in patients with hypercupremia (pregnancy, infections, leukemia) the serum copper level correlates closely (correlation coefficient of +0.79) with the  $\alpha_2 + \alpha_3$  globulin fraction of the serum proteins as determined by

TABLE III

*Proteinuria, cupriuria, and siderinuria in patients with the nephrotic syndrome\**

Patient	Plasma		No. of determinations	Urine				
	Copper	Iron		Protein	Copper	Cu P	Iron	Fe P
	$\mu\text{g./100 ml.}$	$\mu\text{g./100 ml.}$						
C. E.	63	33	10	8.8	270	31	466	53
R. K.	58	64	12	15.4	490	32	864	56
G. H.	94	95	11	12.2	346	28	610	50
W. B.	89	96	11	5.6	183	33		
D. G.	49	42	6	18.4	324	18	453	25
E. C.	20	117	3	9.3	92	10	480	52

\* The ratios are expressed in  $\mu\text{g.}$  of copper (Cu) or iron (Fe) per gram of protein (P).

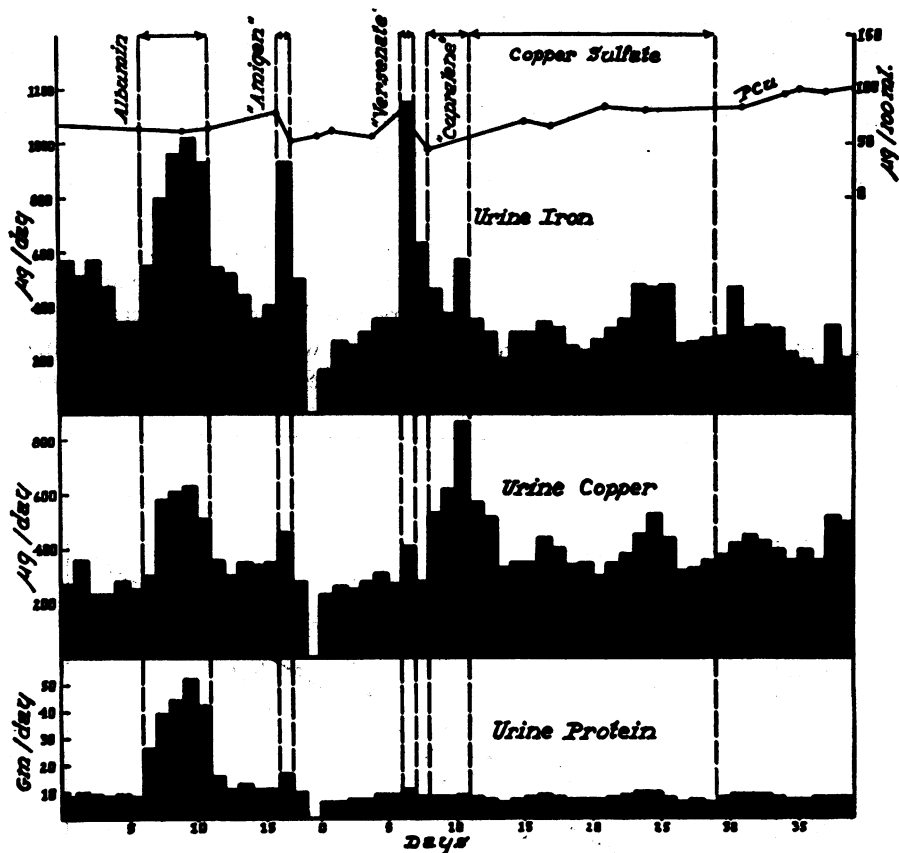


FIG. 3. THE INFLUENCE OF THE ADMINISTRATION OF ALBUMIN, "AMIGEN," "CALCIUM VERSENATE," "CUPRALENE," AND COPPER SULFATE ON THE URINARY EXCRETION OF PROTEIN, COPPER, AND IRON AND ON THE PLASMA COPPER LEVEL IN A PATIENT (C. E.) WITH THE NEPHROTIC SYNDROME

Fifty grams of human serum albumin were given daily for five days. One liter of casein hydrolysate ("Amigen") was next given intravenously. This was followed by a single intravenous injection of one gram of the calcium disodium salt of ethylenediamine tetraacetic acid ("Versenate"). For three consecutive days, 100 mg. of allylcuprothiocarbamate ("Cupralene") was given intravenously each day. Finally copper sulfate was administered by mouth in an amount of 33 mg. three times a day for a period of 18 days.

electrophoretic analysis. Since it is known that the serum of patients with the nephrotic syndrome contains increased amounts of  $\alpha_2 + \alpha_3$  globulins (23), it is of interest that in this condition no such correlation was observed (Figure 1).

In an effort to study the disappearance rate of intravenously administered copper, 100 mg. of "Cupralene" (allylcuprothiocarbamide, containing 19.93 per cent copper) were administered to a patient with the nephrotic syndrome and the results compared with the disappearance rate in a normal subject (Figure 2). In the nephrotic patient the starting value was lower but the magnitude of increase was comparable to the increase in the normal subject and the rate of disappearance from the plasma was the same in both individuals. In both the normal subject and the patient with the nephrotic syndrome, within the limits of experimental error all of the injected copper appeared in the direct-reacting fraction of the plasma copper (11).

#### *Urinary copper, iron and protein in the nephrotic syndrome*

Mean values for the excretion of copper, iron, and protein in the urine of six patients are pre-

sented in Table III. The plasma values for copper and iron are also given.

The patients excreted from 92 to 490  $\mu\text{g.}$  of copper per day. In the first four patients (C. E., R. K., G. H., and W. B.) the copper/protein ratio was extremely constant. Approximately 31  $\mu\text{g.}$  of copper were excreted per gram of protein. Prior to the administration of various substances this ratio was relatively constant from day to day in each of the six adult patients. This is illustrated in a single patient (C. E.) in Figure 3. The ratio was significantly lower (18 and 10) in patients D. G. and E. C. It is of interest that the plasma copper levels in these two patients were extremely low.

In Figure 4, the daily urinary protein values are plotted against the daily values for the excretion of copper in the first four patients of Table III. The mean value for ten normal subjects (Table I) is also plotted. The correlation coefficient ( $r$ ) for these values is +0.97. Only the mean values for the determinations on E. C. and D. G., have been plotted. In both of these patients, as noted above, there was less copper excreted per gram of protein than was excreted in the other four patients.

The excretion of iron was somewhat greater

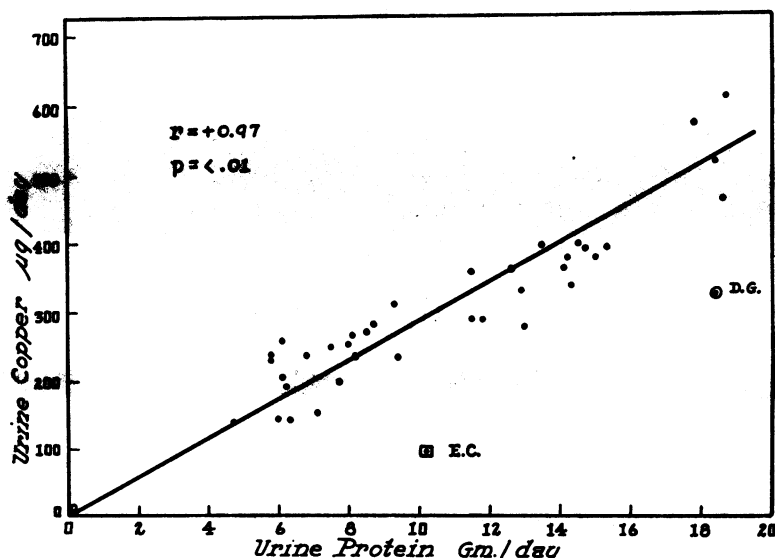


FIG. 4. THE CORRELATION BETWEEN THE DAILY URINARY EXCRETION OF COPPER AND PROTEIN IN FOUR PATIENTS (C. E., R. K., G. H., AND W. B.) WITH THE NEPHROTIC SYNDROME

The mean value for 10 normal subjects (open dot, lower left hand corner) and the mean values for patients D. G. and E. C. are also given. The line was plotted by the method of least squares.

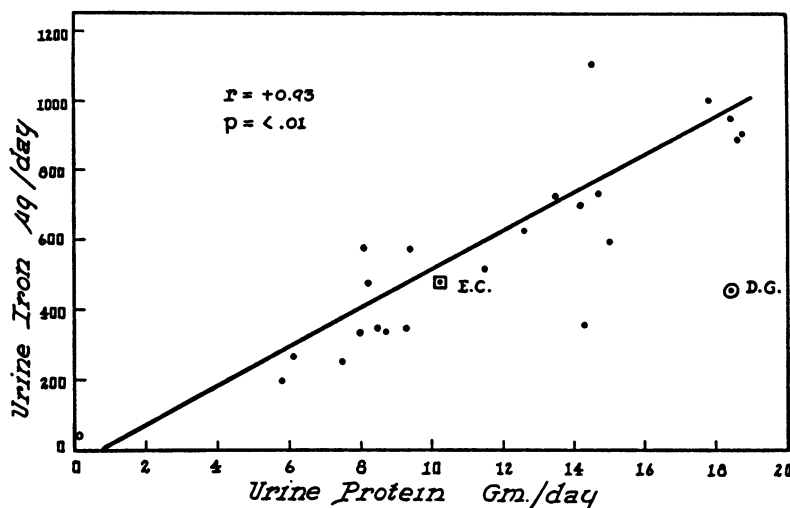


FIG. 5. THE CORRELATION BETWEEN THE DAILY URINARY EXCRETION OF IRON AND PROTEIN IN THREE PATIENTS (C. E., R. K., AND G. H.) WITH THE NEPHROTIC SYNDROME

The mean values for 10 normal subjects (open dot) and the mean values for D. G. and E. C. are also given. The line was plotted by the method of least squares.

than that of copper and ranged from 453 to 864  $\mu\text{g.}$  per day. The ratio iron/protein varied from 50 to 60 in four of the patients but was only 25 in one of the patients (D. G.). In this patient the plasma iron was low. However, it should be noted that in patient C. E. the plasma iron was equally low and yet the urinary iron/protein ratio was 53. The iron/protein ratio was relatively constant from day to day in patient C. E., as illustrated in Figure 3.

In Figure 5, the daily urinary protein excretion is plotted against the daily iron excretion for three of the patients (C. E., R. K., and G. H.). The mean value for the 10 normal subjects (Table I) is also shown. The correlation coefficient ( $r$ ) for

these data is +0.93. Only the mean values for E. C. and D. G. have been plotted. The mean value for E. C. fell on the line. In the case of D. G., as noted above, the mean value fell below the line.

Specimens of urine from all six of the patients were dialyzed repeatedly against distilled water for 18 hours at  $2^\circ\text{C.}$  In no instance was a detectable quantity of the copper or iron dialyzable.

In order to study the influence of various substances on the urinary copper/protein and iron/protein ratios, human serum albumin, casein hydrolysate ("Amigen"), allylcuprothiocarbamide ("Cupralene"), and the calcium disodium salt of ethylenediamine tetra-acetic acid ("Calcium Versenate") were given intravenously, and copper

TABLE IV  
*Proteinuria, cupriuria, and siderinuria during the intravenous administration of albumin\**

Patient	Period	Protein	Copper	Cu P	Iron	Fe P
		Gm./24 hrs.	$\mu\text{g.}/24\text{ hrs.}$		$\mu\text{g.}/24\text{ hrs.}$	
C. E.	Before	8.8	270	31	466	53
	During	52.2	625	12	1020	19
G. H.	Before	12.2	346	28	610	50
	During	38.6	464	12	1060	27
G. H.	Before	8.4	225	27	391	47
	During	21.5	351	16	643	30

\* The ratios are expressed in  $\mu\text{g.}$  of copper (Cu) or iron (Fe) per gram of protein (P).



TABLE V  
*Proteinuria, cupriuria, and siderinuria during the intravenous administration of casein hydrolysate \**

Patient	Period	Protein	Copper	$\frac{\text{Cu}}{\text{P}}$	Iron	$\frac{\text{Fe}}{\text{P}}$
		Gm./24 hrs.	$\mu\text{g.}/24 \text{ hrs.}$		$\mu\text{g.}/24 \text{ hrs.}$	
C. E.	Before	11.6	336	29	426	38
	During	16.6	514	31	934	56
R. K.	Before	17.1	489	29	863	50
	During	16.3	569	35	1184	73
D. G.	Before	19.8	314	16	444	22
	During	20.3	419	21	1000	49

\* The ratios are expressed in  $\mu\text{g.}$  of copper (Cu) or iron (Fe) per gram of protein (P).

sulfate was given orally to each of several patients.

The relationship of intravenously administered human serum albumin to the urinary copper/protein and iron/protein ratios is summarized in Table IV. C. E. was given 50 grams of albumin daily for four days. G. H. on two separate occasions received 25 grams daily for two consecutive days. In each case, the administration of albumin was associated with a significant increase in the excretion of both copper and iron. However, the increase in urinary total protein was considerably greater, with the result that there was a significant decrease in the copper/protein and iron/protein ratios.

The intravenous administration of one liter of casein hydrolysate (Table V) in each case was associated with a modest but significant increase in the excretion of copper and a somewhat greater diuresis of iron. However, in two of the patients there was a slight increase in the excretion of protein with the result that the copper/protein and iron/protein ratios increased only slightly.

The intravenous injection of "Cupralene" resulted in a moderate to marked increase in the

copper/protein ratio but there was no significant increase in the iron/protein ratio in the two patients in whom the latter was studied (Table VI). In spite of the increase in the copper/protein ratio, only a small amount (0.4 to 2.4 per cent) of the copper injected appeared in the urine during the brief period of study. Patients W. B. and R. K. were given a single intravenous injection of 100 mg. of the compound. Patient C. E. was given 100 mg. daily for three consecutive days.

The influence of the oral administration of copper sulfate on the urinary copper/protein and iron/protein ratios in three patients is summarized in Table VII. Patients W. B. and R. K. were given 33 mg. of copper sulfate ( $5 \text{ H}_2\text{O}$ ) three times a day with meals for five days. Patient C. E. was given the same daily amount of copper for a period of 18 days. In each of the three patients there was a distinct increase in the copper/protein ratio. The iron/protein ratio was not altered significantly.

A single intravenous injection of one gram of "Calcium Versenate" (Table VIII) resulted in no appreciable increase in the excretion of copper in

TABLE VI  
*Proteinuria, cupriuria, and siderinuria during the intravenous administration of "Cupralene" \**

Patient	Period	Protein	Copper	$\frac{\text{Cu}}{\text{P}}$	Per cent copper excreted	Iron	$\frac{\text{Fe}}{\text{P}}$
		Gm./24 hrs.	$\mu\text{g.}/24 \text{ hrs.}$			$\mu\text{g.}/24 \text{ hrs.}$	
W. B.	Before	5.8	106	18	0.4		
	During	5.8	190	32			
C. E.	Before	7.1	252	36	2.4	345	49
	During	8.3	746	90			
R. K.	Before	13.4	403	30	0.9	798	60
	During	13.4	573	43			

\* The ratios are expressed in  $\mu\text{g.}$  of copper (Cu) or iron (Fe) per gram of protein (P).

TABLE VII  
*Proteinuria, cupriuria, and siderinuria during the oral administration of copper sulfate\**

Patient	Period	Protein	Copper	Cu P	Iron	Fe P
		Gm./24 hrs.	µg./24 hrs.		µg./24 hrs.	
W. B.	Before	6.1	114	19		
	During	3.9	131	35		
R. K.	Before	15.8	411	26	864	51
	During	13.2	508	38	744	56
C. E.	Before	7.4	264	36	273	37
	During	6.9	330	48	266	38

\* The ratios are expressed in µg. of copper (Cu) or iron (Fe) per gram of protein (P).

relation to protein but was associated with a striking increase in the excretion of iron.

The influence of albumin, casein hydrolysate, "Calcium Versenate," "Cupralene," and copper sulfate on the excretion of copper, iron, and protein in the urine of one of the patients (C. E.) is illustrated graphically in Figure 3.

#### *Hematologic studies in the nephrotic syndrome*

As shown in Table II, a mild degree of anemia (volume of packed red cells between 30 and 33 ml. per 100 ml.) was present in three of the six adult patients. In a fourth patient (E. C.) a moderately severe anemia existed. However, the condition was somewhat complicated in this patient in that uremia was also present. Anemia of a mild degree was present in five of the ten children and in one the anemia was moderately severe.

The morphologic characteristics of the erythrocytes as well as the percentage of reticulocytes and the leukocyte and platelet counts in the six adult patients are presented in Table IX. In all instances the erythrocytes were normal in size and hemoglobin content. Anemia, when present, was unaccompanied by reticulocytosis or a sig-

nificant alteration in the leukocyte or platelet level.

The changes associated with the administration of copper by mouth are summarized in Table X. Two of the patients (W. B. and R. K.) were given 33 mg. of copper sulfate (5 H<sub>2</sub>O) three times daily with meals for a period of five days. The third patient was given this amount of copper for a period of 18 days. In none of the patients was this therapy followed by a significant reticulocytosis or a significant increase in the volume of packed red cells. In one of the patients (W. B.), at the time this therapy was instituted the plasma copper level was normal. In a second patient (R. K.), the level of copper in the plasma was reduced but did not rise with therapy. In the patient given the longest course of therapy (C. E.), the administration of copper was associated with an increase in the plasma copper level to within the normal range. It should be noted that in this patient the oral therapy was immediately preceded by the intravenous administration of "Cupralene" for three days (Figure 4). In neither of the two patients in whom the plasma iron level was studied, was there any increase in this value following the administration of copper.

TABLE VIII  
*Proteinuria, cupriuria, and siderinuria during the intravenous administration of "Calcium Versenate"\**

Patient	Period	Protein	Copper	Cu P	Iron	Fe P
		Gm./24 hrs.	µg./24 hrs.		µg./24 hrs.	
C. E.	Before	7.4	264	36	273	37
	During	10.9	410	38	1150	105
R. K.	Before	15.9	710	45	786	50
	During	16.1	744	46	1755	109

\* The ratios are expressed in µg. of copper (Cu) or iron (Fe) per gram of protein (P).

TABLE IX  
Hematologic data in the patients with the nephrotic syndrome\*

Patient	V.P.R.C. ml./100 ml.	M.C.V. C $\mu$	M.C.H.C. %	Retics. %	W.B.C. $\times 10^3/mm.^3$	Platelets $\times 10^3/mm.^3$
C. E.	33	82	35	1.8	5.2	184
R. K.	42	88	32	0.8	8.7	280
G. H.	30	80	35	0.5	3.6	300
W. B.	30	88	35	1.4	10.2	320
D. G.	43	89	34	1.0	6.2	230
E. C.	26	88	35	1.4	5.5	240

\* V.P.R.C., volume of packed red cells; M.C.V., mean corpuscular volume; M.C.H.C., mean corpuscular hemoglobin concentration; Retics., reticulocytes; W.B.C., white blood cell count.

#### Plasma copper and iron in patients with uremia

To ascertain if hypocupremia and hypoferremia occur in patients with severe impairment of renal function but without massive proteinuria, the plasma copper and iron levels were studied in seven patients with uremia (Table XI). In none of the patients was hypocupremia observed. In two of the patients hypercupremia was present. A reduction of the plasma iron level was observed in three of the patients. These observations are in accordance with previous studies in this laboratory (4).

#### Plasma copper and plasma iron in hypoalbuminemic states other than the nephrotic syndrome

To determine if hypocupremia and hypoferremia occur in hypoalbuminemic states other than the nephrotic syndrome, studies were made in five patients with cirrhosis of the liver, a patient with nutritional hypoalbuminemia, and a single patient with sprue (Table XII). In none was hypocupremia present and a significant reduction in the plasma iron level was observed in only one patient.

TABLE X  
The influence of the oral administration of copper sulfate on the blood

Patient	Day	V.P.R.C.* ml./100 ml.	Retic. Peak %	Plasma copper $\mu g./100 ml.$	Plasma iron $\mu g./100 ml.$
C. E.	0	34	1.8	57	35
	18	33	3.1	101	30
W. B.	0	31	2.3	115	
	5	31	2.0	106	
R. K.	0	43	1.6	60	66
	5	44	1.6	48	59

\* V.P.R.C., volume of packed red cells.

#### DISCUSSION

From the data presented it is evident that, in at least some patients with the nephrotic syndrome, there is a markedly increased excretion of copper and iron in the urine with a concomitant reduction in the plasma level of these two elements. It seems likely that the hypocupremia and the hypoferremia are, in part at least, the consequence of the loss of ceruloplasmin and transferrin into the urine. For the present, however, this is but an assumption, since ceruloplasmin and transferrin were not isolated and identified in the urine. Nevertheless, it is not unreasonable to think that these two relatively low molecular weight proteins will pass through the kidney in nephrosis.

Several observations indirectly support this view. First, the decrease in the plasma concentration of copper and iron appeared to be associated with a decrease in the concentration of both ceruloplasmin and transferrin. Thus, since approximately 96 per cent of all of the copper in plasma is normally present as ceruloplasmin (1, 11), in all of the patients with a marked reduction in the plasma copper level, it is probable that there was a reduction in the ceruloplasmin concentration. That the concentration of the transferrin was reduced is indicated by the recorded values for total-iron binding capacity of the plasma. Secondly, the observations that neither element could be dialyzed from the urine and that there was a high degree of correlation between the amount of proteinuria and the amount of iron and copper in the urine, indicate that these elements were bound to protein. That such binding to protein was not indiscriminate, however, was indicated by the observation that, when albumin was administered intravenously there was a marked reduction in the ratio of copper

TABLE XI  
*Plasma copper and iron in patients with uremia*

Patient	V.P.R.C.*	BUN*	Total plasma protein	Plasma albumin	Plasma globulin	Plasma copper	Plasma iron
	<i>ml./100 ml.</i>	<i>mg./100 ml.</i>	<i>Gm./100 ml.</i>	<i>Gm./100 ml.</i>	<i>Gm./100 ml.</i>	<i>µg./100 ml.</i>	<i>µg./100 ml.</i>
D. T.	26	114	6.2	3.6	2.6	101	82
V. L.	23	190	6.2	5.0	1.2	119	162
D. H.	24	129	5.6	3.9	1.7	181	63
L. F.	26	80	4.8	2.7	2.1	149	94
V. J.	25	204	6.6	2.9	3.7	161	59
F. L.	36	129	7.2	4.0	3.2	134	102
M. L.	32	148	4.5	2.5	2.0	104	36

\* V.P.R.C., volume of packed red cells; BUN, blood urea nitrogen.

and iron to protein in the urine. This would suggest that copper and iron were not being excreted bound to albumin. The observed increase in the absolute amount of excreted copper and iron could be explained by a greater loss of ceruloplasmin and transferrin secondary to the increased glomerular filtration rate which is known to occur following the intravenous administration of albumin (24). Finally, if copper and iron were to be excreted other than as ceruloplasmin and transferrin, a renal mechanism for the removal of copper and iron from these proteins and for combining them with other proteins would have to be postulated. Obviously, there is no such mechanism present in the normal kidney since, at most, only negligible amounts of these elements are excreted in the urine. It seems unlikely that such a mechanism would be present in the kidneys of patients with the nephrotic syndrome.

However, if the hypocupremia and hypoferrinemia are to be explained solely on the basis of the loss of ceruloplasmin and transferrin in the urine, it then follows that the degree of reduction of these two proteins in the plasma, assuming the initial stores to be equal and the rate of synthesis to be

constant, should depend upon the duration of the nephrotic state and the amounts of these two proteins lost in the urine. In this study, in a small group of patients observed for a comparatively short period of time there was no apparent correlation between the duration of the syndrome and the plasma copper and iron levels. Likewise, there was no correlation in six patients between the degree of cupriuria or siderinuria and the degree of hypocupremia or hypoferrinemia. In two of the patients (G. H. and W. B.) with plasma copper values within the normal range, and in two of the patients (G. H. and E. C.) with normal plasma iron values, there was a marked degree of cupriuria and siderinuria. Admittedly, if a larger group of patients were to be studied through the entire course of their illness, such a correlation might be found. At this point it can only be stated that the limited data available suggest that the levels of copper and iron in the plasma depend upon other factors in addition to the degree of urinary loss.

Hypercupremia is known to be present in patients with infections (25). In the patients with the nephrotic syndrome with recognized infections there was a tendency for the hypocupremia to be

TABLE XII  
*Plasma copper and iron in hypoalbuminemic states other than the nephrotic syndrome*

Patients	Diagnosis	Total plasma protein	Plasma albumin	Plasma globulin	Plasma copper	Plasma iron
		<i>Gm./100 ml.</i>	<i>Gm./100 ml.</i>	<i>Gm./100 ml.</i>	<i>µg./100 ml.</i>	<i>µg./100 ml.</i>
R. V.	Cirrhosis of liver	7.0	2.6	4.4	156	126
R. Q.	Cirrhosis of liver	5.8	2.5	3.3	133	116
R. P.	Cirrhosis of liver	6.7	2.3	4.4	101	135
A. M.	Cirrhosis of liver	5.9	2.0	3.9	159	87
B. H.	Cirrhosis of liver	5.5	2.1	3.4	132	110
L. L.	Nutritional hypoalbuminemia	6.0	2.1	3.9	155	56
M. M.	Sprue	4.8	2.9	1.9	113	22

less severe than in those without infections but this correlation did not hold in all of the cases and a severe degree of hypocupremia was present in several patients with complicating infections. It is possible that in several of the patients infection contributed to the reduction in the plasma iron level, but a recognized complicating infection of mild nature was present in the four patients with a normal plasma iron level and in the other patients hypoferremia existed in the absence of a recognized infection.

In addition to the above, the fact that hypocupremia and hypoferremia are not present in all patients with the nephrotic state, even when there is considerable loss of copper and iron in the urine, suggests the possibility that in such patients there may be considerable variation in the ability of the body to synthesize these two metal-containing proteins. Severe hypocupremia and hypoferremia may develop only when urinary loss is accompanied by an impaired rate of synthesis.

In the two patients (D. G. and E. C.) with the lowest plasma copper levels, the ratio of copper to protein in the urine was considerably lower than in the other four with higher plasma copper values. It may be that when the plasma copper level falls below a certain value, the rate of loss in the urine diminishes.

The intravenous administration of casein hydrolysate was associated with a definite increase in the amount of copper and iron in the urine. In all three instances this increase was greater than the increase in proteinuria. This suggests that a small amount of copper and iron may have been excreted as a complex with aminoacids. However, none of the copper or iron was dialyzable from the urine.

We have no information concerning the degree of depletion of copper and iron in the body of patients with the nephrotic syndrome. It would be desirable, if the occasion should present itself, to measure the copper and iron content of the tissues, particularly the liver and spleen. The finding of a low content of copper in the erythrocytes of three of the five patients in whom this determination was performed suggests that in at least certain of the tissues of some patients there may be a significant depletion of copper. In the copper-deficient swine (26, 27) it has been found that the

concentration of copper in the erythrocytes does not decline until the deficiency becomes severe.

Since severe iron deficiency in human subjects is manifested by a microcytic, hypochromic anemia, and copper-deficiency in swine is accompanied by a similar type of anemia (26), it would seem unlikely that depletion of these two elements is the explanation for the normocytic, normochromic type of anemia which accompanies the nephrotic state in human subjects. This is also suggested by the observations reported here, limited though they were, in which the anemia failed to respond to the administration of copper.

In a recent comprehensive study (5) of the plasma copper level in approximately 200 patients with a variety of diseases, it was pointed out that hypocupremia is an extremely uncommon finding. To date, hypocupremia has been observed consistently in human subjects in only three situations; namely, the newborn (28), Wilson's disease (5, 29, 30), and the nephrotic syndrome. A dietary deficiency of copper has not, as yet, been clearly demonstrated in a human subject (31).

#### SUMMARY

Hypocupremia was observed in 13 of 16 patients with the nephrotic syndrome and hypoferremia in 10 of the 16. The mean plasma copper level in the 16 patients was  $64 \pm 20$   $\mu\text{g.}$  per 100 ml. as compared with  $116 \pm 14$   $\mu\text{g.}$  per 100 ml. in normal subjects. The mean plasma iron value was  $50 \pm 33$   $\mu\text{g.}$  per 100 ml. as compared with the normal mean of  $110 \pm 31$   $\mu\text{g.}$  per 100 ml. The total iron-binding capacity of the plasma, determined in seven patients, was found to be reduced in all seven. The mean value was 78  $\mu\text{g.}$  per 100 ml. as compared with the normal mean of 359.

Patients with the nephrotic syndrome were found to excrete increased quantities of copper and iron in the urine. The copper and iron were not dialyzable and the amount excreted was correlated with the amount of protein present in the urine. Approximately 31  $\mu\text{g.}$  of copper and 53  $\mu\text{g.}$  of iron were excreted per gram of protein. However, following the intravenous administration of albumin there was a marked increase in the degree of proteinuria and but only a slight increase in the amount of copper and iron excreted.

Hypocupremia was not observed in patients

with uremia or hypoalbuminemic states other than the nephrotic syndrome. Hypoferremia was observed in a few patients with these conditions.

It is suggested that the hypocupremia and the hypoferremia associated with the nephrotic state are, in part at least, the consequence of the loss of ceruloplasmin and transferrin in the urine. However, the possibility that there is impairment in the rate of synthesis of these two metal-binding proteins in some patients cannot be ruled out.

Red blood cell copper was found to be reduced in three of five patients in whom this determination was made. The anemia failed to respond to the administration of copper in a limited number of patients in whom this was studied.

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