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CARBON DIOXIDE AND OXYGEN TENSIONS OF THE MIXED VENOUS BLOOD OF MAN AT REST

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PART I

THE PROBLEM OF EQUILIBRATION OF LUNG GASES WITH INCOMING VENOUS BLOOD

The state of the mixed venous blood, or the blood in the right side of the heart and in the pulmonary artery, especially in respect to its oxygen and carbon dioxide contents, has been of interest since Fick's principle (1) of measuring circulation rate was enunciated in 1870.

In animals the study of this blood can be made directly by puncture of the right side of the heart or by insertion of canulae through the jugular vein into the right auricle. In man such procedures have not usually been considered justified for purposes of experimental study, and recourse has therefore been had to the use of the lungs as an aërotonometer, in the attempt to bring the lung gases into equilibrium with inflowing venous (pulmonary artery) blood before recirculation alters the character of this inflowing blood. Application of these tensions to oxygen or carbon dioxide dissociation curves of the blood then gives the O₂ and CO₂ contents of the mixed venous blood. Numerous methods on this principle have been devised.

Very recently some experiments have been performed with human subjects, in which blood from the right heart was obtained directly. These will be referred to later.

Loewy and von Schrötter (2) in 1905 performed some experiments in which they passed a catheter down the subject's trachea and into one of the large bronchi; then closed off the bronchus by inflating a small rubber bag around the catheter. This portion of the lung thus acted as a closed tonometer. When sufficient time had elapsed in order for

equilibrium to be reached, a sample of the air was withdrawn through the catheter. The sample was supposed to have the same tensions of O_2 and CO_2 as the incoming venous blood. This method assumed that the procedure did not appreciably affect the general circulo-respiratory equilibrium; and it also called for considerable heroism on the part of the subject. It has been little used since.

Four years later Plesch (3) developed the technique of rebreathing gas mixtures in and out of a rubber bag. He attempted to bring the rebreathed mixture into equilibrium both with the O_2 and CO_2 of the venous blood. In order to make the oxygen of the mixture sufficiently low, a preliminary rebreathing of pure nitrogen was employed. Plesch was the first to make the assumption that if, after successive periods of rebreathing, the gases in the bag reached constant values, these values therefore represented equilibrium with the incoming venous blood. We shall return later to a consideration of this assumption.

The measurement in 1914 by Christiansen, Douglas, and Haldane (4) of the effect of oxygenation upon the CO_2 dissociation curves of whole blood led to a further simplification of technique. If a proper mixture of CO_2 and air (or CO_2 and O_2) were rebreathed, it should be possible to oxygenate the incoming blood, without at the same time causing any alteration in its CO_2 content. The CO_2 tension of this blood will, of course, be raised over the true venous level, due to its oxygenation. Blood in this state was called "oxygenated venous," or (Y. Henderson and Prince (8)) "virtual venous" blood. The technique of Christiansen, Douglas, and Haldane was as follows: the subject, after a short preliminary rest period, inhaled from a small rubber bag a mixture of from 6.5 to 10.0 per cent CO_2 in air (sometimes CO_2 in oxygen), held the mixture in the lungs $4\frac{1}{2}$ seconds, then expired about 1 liter, held the breath 6 seconds longer, then made a complete expiration. Samples of the two fractions were analyzed. In some experiments three partial expirations were made, at 4, 9, and 13 seconds. If the CO_2 values of these successive samples were the same, or nearly so, it was believed that equilibrium had been obtained. Further evidence of equilibrium was obtained by having the subject rebreathe successively (*a*) a mixture which when mixed with alveolar air would give a CO_2 tension higher than the oxygenated mixed venous; and then (*b*) a mixture giving too low a tension. Thus the equilibrium level was ap-

proached both from above and below. It was assumed here that the lung gases became completely mixed by being simply held in the lungs for $4\frac{1}{2}$ seconds; and also that the equilibrium with the incoming blood could be reached in $4\frac{1}{2}$ (or 9) seconds by the necessary transfer of gases to or from the circulating pulmonary blood. Theoretically, if mixture in the lungs were not perfect, it would have been quite possible for an inhaled mixture that was too low to give in the $4\frac{1}{2}$ second sample a value higher than the completely mixed air, on account of the greater proportion in that sample of air from the bag; while the $10\frac{1}{2}$ second sample might equal this value by increase of CO_2 in it, derived from the blood; without, however, complete equilibrium with the venous blood having been reached at any time. In the cases where successive experiments with different initial mixtures were performed, and good agreement was obtained, this possibility was eliminated.

The results of Christiansen, Douglas, and Haldane showed rather large variations in equilibrium values obtained, between successive experiments, also at times in single experiments agreement of samples was not good. However, the general level of oxygenated mixed venous tension was found, and this marked at the time a signal advance.

The calculations whereby Christiansen, Douglas and Haldane, transferred tensions into terms of blood gas contents were cumbersome, and involved the calculation of the actual mixed venous CO_2 tensions.

Fridericia in 1918 (5) determined venous CO_2 tensions by a technique similar to that of Christiansen, Douglas, and Haldane, except that when differences were found between the tensions obtained at half and full expiration, he calculated the amounts of CO_2 absorbed or given off, and by using varying initial mixtures was able to estimate the tension of zero absorption, or equilibrium. By a method slightly modified from this, Liljestrand and Lindhard in 1920 (6), compared circulation rates as calculated by this and by the Krogh-Lindhard (7) nitrous oxide method, and found satisfactory agreement.

Henderson and Prince (8) in 1917 developed a method for measuring venous tensions similar to that of Plesch with the difference that CO_2 alone was measured. The method consisted in making a deep expiration into a bag, then, after a rest interval of several minutes, inhaling this mixture, holding it in the lungs for 5 to 10 seconds, and exhaling it again; this procedure was repeated several times, and it was found that

a CO_2 tension was reached after 3 or 4 trials which remained constant through further rebreathings. This was considered to be the oxygenated venous CO_2 tension, and it was suggested that the venous CO_2 content could then be obtained by equilibrating a sample of the subject's blood, with this rebreathed air. Henderson and Prince checked their method in two ways: (a) they found that by their method of "intermittent rebreathings" the same final equilibrium tensions were obtained whether they started with a high or low CO_2 tension in the rebreathing bag. They also investigated the effects of holding the air in the lungs (or of rebreathing it slowly) for varying lengths of time, from 3 to 20 seconds or more. Thus, they obtained one equilibrium value, with successive periods of rebreathing, each of 3 seconds' duration, another value after a series of rebreathing, each of 10 seconds' duration, and so forth. They found that between 6 and 16 seconds, a "plateau" level was reached, the equilibrium values obtained for these durations of rebreathing, being essentially the same; these values they interpreted to be the oxygenated venous CO_2 tension. This was a fair inference, except that they did not know what their oxygen tensions were; and it is possible that the transient interruption in a steady rise of venous CO_2 (as represented by the "plateau") was only an apparent equilibrium, due to a falling off of lung O_2 tension with consequent check in the CO_2 tension rise.

Meakins and Davies (9) in 1922 employed a technique similar to that of Henderson and Prince with the difference that the air was regularly rebreathed in and out of the bag twice, instead of simply being held in the lungs. They believed that in this manner, more perfect mixture could be obtained.

Barcroft, Roughton, and Shoji (10) equilibrated the oxygen in the subject's lungs with that in the venous blood, instead of equilibrating the CO_2 . This they did by simply rebreathing pure nitrogen. They found in successive trials that the oxygen tension soon reached a constant level each time at the end of 7 or 8 seconds of rebreathing. This level was maintained till about the twentieth second, after which it decreased rapidly. (This latter fact demonstrated the time of onset of recirculation of blood.) The CO_2 tension at this equilibrium level was lower than that of true mixed venous blood, on account of the

large admixture of nitrogen and very low oxygen tension. Barcroft and his associates recognized this and made certain corrections for it.

At about the same time, Redfield, Bock and Meakins (11) devised an extrapolation method for determining simultaneously both the actual CO_2 and O_2 tensions of mixed venous blood. They rebreathed a given mixture of CO_2 , O_2 and N_2 , and took a sample at the end of 5 seconds and another at the end of 10 seconds. These were analyzed, plotted on a CO_2 — O_2 tension diagram and a line drawn, connecting the two and extrapolated beyond. Similar experiments were done using different initial rebreathed mixtures. It was found that the lines drawn through the various pairs of 5 and 10 second points, when extrapolated, tended to meet at the same point. This was inferred to represent the mixed venous tensions. The procedure assumed that the rate of change of CO_2 , during equilibration, held a constant ratio to that of O_2 . The authors found further that the point of intersection of two or more extrapolated lines was also crossed by a horizontal line drawn at the CO_2 tension as obtained by the Henderson-Prince technique. This last point casts a certain doubt upon the accuracy of the method; because the Henderson-Prince *oxygenated* venous CO_2 tension ought to have been considerably above their own *true* mixed venous CO_2 tension.

Recently Grollman (12) has compared values of cardiac output as obtained by the use of this extrapolation technique, with similar values by the acetylene method, and finds satisfactory agreement.

In 1922 Douglas and Haldane (13) continued further the experiments which Christiansen, Douglas, and Haldane had made in 1914. Knowing the oxygenated and reduced CO_2 dissociation curves and the O_2 dissociation curves of their subjects, these workers were able to find both the oxygenated mixed venous CO_2 levels and the true mixed venous CO_2 and O_2 levels in successive experiments on the same day, and compare the arteriovenous differences (or circulation rates), so obtained. The method used differed slightly from their earlier one. A mixture was made up in a Douglas bag which had approximately the desired concentrations of CO_2 and O_2 (previously determined by a pilot experiment). Three successive complete inspirations were then made from the bag, each inspiration having been preceded by a complete expiration to the outside air. The last inspiration was held

two seconds, a half expiration made, a sample taken, then after five seconds more a complete expiration made and a second sample taken. If the gas tensions in the two samples were the same, it was concluded that equilibrium had been reached. If not, then the equilibrium tension was obtained by extrapolation, by adding (algebraically) to the second tension the difference between the first and second. This procedure was admittedly empirical but gave consistent results and was therefore accepted.

The results of Douglas and Haldane on several subjects, both at rest and during work, showed essentially the same arteriovenous differences, and therefore the same mixed venous blood contents of O_2 and CO_2 , whether oxygenated or true venous CO_2 tensions, or venous oxygen tensions, were used.

In recent months, Ewig and Hinsberg (14) have described a method, similar to that of Douglas and Haldane, except that they purposely employ mixtures which give final values on either side of the equilibrium, and calculate the oxygenated venous tension by interpolation.

The technique used by Eppinger, Papp, and Schwarz (38) was somewhat similar to that of Douglas and Haldane, true venous tensions of CO_2 and O_2 being obtained.

Murray and Taylor (15) in a brief communication described another method of reaching equilibrium simultaneously with both CO_2 and O_2 . In this method, a mixture of 1.5 per cent O_2 , 6 per cent CO_2 and 92.5 per cent N_2 was rebreathed, instantaneous samples taken into evacuated tubes at the end of each expiration, and the resulting CO_2 and O_2 tension values plotted on a CO_2-O_2 tension diagram. At equilibrium, successive samples fell on the same point on the chart. The feature of this method which one may emphasize is that the course of equilibration of each *particular* experiment was followed by the use of successive instantaneous samples. Christiansen, Douglas, and Haldane had accomplished this also but took fewer samples.

The method of Field, Bock, Gildea, and Lathrop (16) represented a slight modification of that of Meakins and Davies, in that (a) the number of rebreathings was increased to four or five, and (b) the mixture used was 6 per cent CO_2 and 94 per cent O_2 , instead of CO_2 and air. The authors showed that with the use of the latter, the O_2 tension at the end of several rebreathings was reduced to 65–75 mm.—insuffi-

cient to oxidize the blood completely, and that the CO_2 tension was correspondingly 1 to 3 mm. lower than when $\text{CO}_2\text{--O}_2$ mixtures were rebreathed. They showed also that the same CO_2 level was reached at the end of the second as at the end of the sixth rebreathing, thus indicating a plateau level, from which it was inferred that a state of equilibrium had been reached. The matter of the diluting effect of the residual alveolar air was discussed, as to whether this might prevent a true equilibrium with venous blood being attained. This was tested by adding to the bag 12 cc. of pure CO_2 between rebreathing periods; it was shown in three experiments that the same level was reached in 15–20 seconds, whether the extra CO_2 was added or not. In their further experiments, however, in measuring circulation rates in various individuals the mixed venous values were so variable in some instances, and the circulation rates so large, that the question does arise as to whether a true equilibrium with venous blood can be reached by this method in all subjects.

In further work from the same laboratory, recently reported, (17) this possible source of error has been eliminated. The technique now employed is to have the mixture in the rebreathing bag about 2 or 3 mm. higher in CO_2 content than the expected oxygenated venous level, thus compensating for the diluting effect of the alveolar air. The present authors also, in a recent communication (18), have reported a similar technique.

Field, Bock, Gildea, and Lathrop confirmed Barcroft, Roughton, and Shoji in demonstrating that at rest, the time when evidences of recirculating blood began to be apparent was at the end of about 20 seconds of rebreathing, the CO_2 tension then starting a progressive rise above the plateau level.

The importance of having sufficient oxygen in the rebreathing mixture, in all methods of obtaining oxygenated venous tensions, is stressed in two recent papers by Israëls and Lamb (19).

The most elaborate of all experiments so far reported for arriving at the actual state of the mixed venous blood have been those of Burwell and Robinson (20). These investigators by the use of a double rebreathing system, first washed out the subject's lungs by two respirations from a spirometer containing nitrogen, then completed the equilibration by rebreathing suitable mixtures of CO_2 , O_2 , and N_2 .

They took their sample at the end of about 20 seconds from the time of the first respiration. By employing this technique successively with mixtures that were above and below equilibrium levels, in their initial CO_2 and O_2 tensions, and reaching essentially the same final values with each, they concluded that they had arrived at the true mixed venous tensions. In order to convert tensions into blood gas contents, samples of venous blood were equilibrated in tonometers with the lung air samples obtained at the end of rebreathing.

This method, though laborious, would certainly seem to be as nearly free of theoretical objections as any that has yet been described.

One further point has received special attention recently and that is the question of completeness of mixture of the lung gases,—whether the samples in the rebreathing bag are identical with the air in the pulmonary alveoli. On this subject, Grollman and Marshall (21) have reported some experiments which indicate that only after about 5 rebreathings taking at least 12 to 15 seconds, is such complete mixture obtained. On the other hand, Bock, Dill and Talbott (17) have found that simply holding an inhaled mixture in the lungs for 5 second periods is as efficient in this respect as rebreathing the mixture a greater number of times. Earlier work on this problem was done by Sonne (22).

Hamilton, Moore and Kinsman (23), in a recent paper, have justly criticized the intermittent rebreathing methods, in that in all of these (except where CO_2 is added to the bag between rebreathings), irrespective of what the initial rebreathed mixture is, actual equilibrium will always be arrived at from below, on account of the diluting effect of the residual air in the lungs. They have, however, advanced a further hypothesis, namely, that at any given moment, while the body is at rest, the "blood in the lungs is not wholly venous but more or less arterial in some parts and venous in others, and the tension of the lung air is an expression of the average of these various tensions." On this hypothesis, they believe that it is possible that the continued rebreathing of a CO_2 — O_2 mixture will only very gradually—in 15 to 20 seconds or more—come into equilibrium with the actual venous (pulmonary artery) blood, being delayed by the arterialized blood in the lungs. We do not see the justification for this latter deduction. Irrespective of how the blood in various individual lung capillaries

may differ in rate of flow, the blood in these capillaries will be exposed at once to pulmonary gases and will presumably come into equilibrium with these gases promptly; whereas all the blood behind this capillary blood will be mixed venous blood, and all blood in front of the capillary blood will be arterialized blood. If the lung and blood tensions are sufficiently close so that lung-blood equilibrium occurs when blood and gas come in contact in the alveoli, then the only way by which a true general equilibrium can be delayed is by incomplete mixing of gases in various parts of the lungs during the first few seconds of re-breathing. If this were the case, there might be some of the alveolar spaces that remained for these few seconds untouched by the re-breathed $\text{CO}_2\text{-O}_2$ mixture. Practically this is not a problem with normal individuals, as by proper re-breathing methods, complete mixture of lung gases can readily be obtained.

The experiments reported by Hamilton, Moore and Kinsman are clear. They used a mixture of 6 per cent of CO_2 in oxygen in a re-breathing bag. Their subject then re-breathed this mixture for six seconds, rested, re-breathed again for six seconds, and so forth until a constant CO_2 tension was found in the bag upon successive re-breathings. The same procedure was employed with 9, 12, 16, and 25 second re-breathing periods. It was found (contrary to Henderson and Prince's work) that no plateau level was reached but that the CO_2 tension increased progressively from the sixth to the twenty-fourth second.

This can readily be explained, it seems to us, by the simple hypothesis that in this subject the diluting effect of the residual air is more than can be made up for by passage of CO_2 from blood into lungs before re-circulation occurs. Another set of experiments is also reported, in which CO_2 was added to the bag before each re-breathing. Again no plateau level was found; but here only three re-breathing times were measured, 8, 16, and 24 seconds, so that even if a plateau had existed in the interval from 10 to 20 seconds (where it has usually been reported) the data would not have shown it.

Further evidence of the "partial arterIALIZATION" of lung blood, as given by the authors, was that upon re-breathing a $\text{CO}_2\text{-O}_2$ mixture, the total CO_2 in lungs and bag together, appeared to decrease during the first few seconds. This CO_2 , it was argued was absorbed by the

arterialized blood in the lungs. It seems more probable, as suggested in a somewhat different connection several years ago by Christiansen, Douglas, and Haldane (4), that the CO_2 thus lost passes into the tissue fluids of the lung, in raising them from alveolar to oxygenated venous CO_2 tension levels.

Mixed venous values can of course be derived indirectly from cardiac output figures as obtained by foreign-gas inhalation methods, providing the gas contents of the arterial blood, and the minute output of CO_2 or absorption of O_2 , are known. The extensive work that has been done by these methods is outside the scope of the present study.

It is self-evident and well-recognized that all the methods cited above, of obtaining mixed venous blood values, are based on a number of inferences.

The development of a method whereby blood can be drawn from the right heart or pulmonary artery, and compared with "mixed venous" values obtained on the given subject at the same time by one of the indirect methods, will probably be necessary before the absolute accuracy of these methods can be known.

Forssmann (24) recently reported an experiment in which he passed a catheter into his own arm vein, along the subclavian and superior vena cava and so into the right heart. He did not make any blood-gas measurements.

Lauter (25), with Bauman and Friedländer, has investigated the volume output of the heart by actual puncture of the right ventricle and of an artery in man, and has found that the average arteriovenous oxygen difference under basal conditions in normal man is about 5 volumes per cent, and the average volume output of the heart about 4.5 liters per minute. This agrees satisfactorily with the results obtained by several of the indirect methods in the literature cited above. In general, certain foreign-gas inhalation methods appear to give values slightly lower than this, and certain Fick principle methods values somewhat higher; this work has been recently reviewed by Marshall (26).

Even after the direct determination of the CO_2 and O_2 in the blood from the right heart has been made, with the establishment of the accuracy, or the amount of error, of the indirect methods, these latter methods will probably continue to be used in experimental work, as

one would expect that the obtaining of blood from the right heart will remain a somewhat formidable procedure.

It is of interest, therefore, to examine these methods as critically as possible, from the point of view of the assumptions implied in accepting their results.

The following are some of the more important assumptions usually made, explicitly or implicitly, in determinations of mixed venous values.

1. The actual procedure of rebreathing, etc., has no appreciable effect on the values obtained. This is not a simple question to decide. In the methods where several deep respirations are employed within a few seconds, the increased respiratory activity will increase the CO_2 tension and decrease the O_2 tension in the blood coming from these muscles; the "bellows" action of the chest will increase the rate of venous blood flow toward the heart, especially from the abdominal viscera. Whether this latter action will increase or decrease the "venosity" of mixed venous blood will perhaps depend largely on the state of the abdominal (especially liver) blood as compared to other venous blood.

On the other hand, in the methods where the breath is held for several seconds in inspiration, the effect will be to slow the general venous return, and there will be presumably less change in respiratory muscle activity.

The fact (17) that essentially the same mixed venous CO_2 values are obtained whether the breath is held, or continuous rebreathing is employed, suggests that the resultant of the effects of this procedure does not involve a large error,—not more, probably, than 0.5 mm. Such error as there is will be expected to make the CO_2 tension as measured, higher than the actual resting mixed venous CO_2 tension.

2. There is often assumed to be perfect mixture of all gases in the lungs, and between lungs and rebreathing bag. In earlier work this was taken for granted; more recently, Grollman and Marshall (21), also Bock, Dill and Talbott (17) have investigated this point, as noted above. This later work would seem to answer objections based on the possible disturbing effects due to the "fan-like" character of pulmonary expansion.

3. The assumption is made that within the duration of these re-breathings (variously taken from 12 to 28 seconds) there is no appreciable amount of recirculating blood. The argument is made, reasonably enough, that if there were recirculating blood in progressively increasing quantities, there would be no "plateau" of equilibrium; as will be shown more fully later, however, a considerable amount of recirculating blood will cause only a slight upward trend to the CO_2 plateau, and such a trend is present in most of the reported data. Thus there is good reason to suspect recirculation of the coronary blood. This error would make the mixed venous CO_2 tension, as measured, something less than 0.5 mm. above the actual mixed venous value.

4. It is assumed that the rebreathed air and the blood in the lungs are in perfect equilibrium in respect to CO_2 , and in respect to O_2 also when true mixed venous tensions are being studied. When oxygenated venous values are being sought, it is assumed that equilibrium exists in respect to CO_2 and that the blood hemoglobin is completely saturated with oxygen. This last point was demonstrated for the Field, Bock, et al. (16) method in two experiments by them, involving arterial punctures 20 seconds after rebreathing was begun. More recent work (27) from the same laboratory, however, suggests that even with hyperventilation and high alveolar O_2 values, the arterial blood is rarely more than 97 per cent saturated. If this is the case, then the assumption of complete saturation involves a small error; this will make the oxygenated mixed venous CO_2 tension, as measured, lower than the actual oxygenated mixed venous O_2 tension.

The assumption of lung-blood equilibrium, so far as the true mixed venous blood (both a CO_2 and O_2 equilibrium) is concerned, has not previously been tested.

5. It is assumed that with the use of high oxygen mixtures to obtain oxygenated venous CO_2 values, when a plateau of CO_2 tension is reached, this will represent on the CO_2 curve the true level of CO_2 in the venous blood. There is a clear fallacy here, first pointed out to one of us several years ago by Dr. C. D. Murray. As oxygen is continually being absorbed, the total volume of rebreathed air is diminishing and CO_2 , therefore, becoming concentrated, and the final equilibrium value obtained for CO_2 will be above that of the incoming venous

blood. The extent of this effect will be shown by calculation presently to be in the region of 0.5 mm.

It is not improbable that the effects described in paragraphs 4 and 5 tend to neutralize each other.

6. Most of the methods previously employed depend, for the demonstration of equilibrium values, upon successive rebreathing periods, usually four or more within 15 to 30 minutes. They thus assume a steady state in the subject's circulo-respiratory conditions. With trained subjects the constancy of results obtained would seem ample justification for this.

7. All of the methods based on the general technique of Henderson and Prince make the further assumption already referred to; namely that the diluting effect of the residual air in the lungs can always be compensated for by CO_2 given off from the blood in the pulmonary vessels; and, therefore, that when the same value is reached, on successive rebreatings, this value represents the mixed venous level. That such an assumption is unwarranted has already been indicated. With a given constancy of technique, some constant value would always be reached eventually, whether this value had any relation to the mixed venous blood or not. Thus, suppose the following conditions: (a) air in bag 2,000 cc., containing 44 mm. CO_2 ; (b) residual air 1700 cc. containing 35 mm. CO_2 ; (c) oxygenated mixed venous tension 47 mm. CO_2 ; (d) and the amount of CO_2 that can be given off into the lung air from the blood in 20 seconds as 20 cc. A simple calculation will show that when this bag mixture is rebreathed for 20 seconds, the final "equilibrium" value will always be 44 mm. CO_2 in the bag and lungs. (In this calculation, for purposes of simplification the effect of O_2 absorption during the process is neglected.)

By altering the length of time of rebreathing, as Henderson and Prince (8) did, or by changing the mixture used, between rebreatings, as Field, Bock, et al. (16), and Bock, Dill, and Talbott (17) have done, this possible source of error can be investigated. It should be borne in mind, however, that this method may be applicable to some subjects and not to others. The subject of Hamilton, Moore and Kinsman's (23) experiments, for example, was apparently one of the latter. In general, those subjects with large arteriovenous differences, low circulation rates, and large residual air volumes relative to the volumes

of air in the rebreathing bag, would be unfavorable types for these experimental methods.

8. It is a further assumption of all these methods, except those in which blood is drawn at the time of the experiment, and CO_2 and O_2 dissociation curves constructed, that although only gaseous tensions are measured, these can be translated into blood contents of CO_2 or O_2 , by the use of standard dissociation curves.

It is clear, however, that for the purpose of obtaining arteriovenous differences, it is not necessary to suppose that the dissociation curves used are identical with that of the subject at the moment of rebreathing. In the case of oxygenated venous CO_2 tensions, the slopes of the curves used, over the physiological range, must be the same as those of the subject's blood. In the case of true mixed venous tensions, both the slopes of the CO_2 curves and the "spread" (distance between reduced and oxygenated curves) must be the same. These principles have been well recognized in recent years.

In sum, one can perhaps say that of these eight assumptions, those in paragraphs 1 and 3 are the most likely to be incorrect, and the least likely to be detected. The effect of respiratory movement and of recirculating blood would both act to make the mixed venous CO_2 figures higher and circulation rate figures lower than the actual values; and the errors would be essentially the same by both Fick-principle and foreign-gas inhalation methods. The total error involved is probably less than 10 per cent of the arteriovenous difference.

In spite of this number of assumptions, most of the various methods used have given results for mixed venous tensions that are of the same order of magnitude,—an average level of from 45 mm. to 50 mm. of CO_2 tension for most subjects. It is also true, however, that some methods, when used for the determination of circulation rate, give definitely higher values than other methods; some methods also give unexpectedly large variations in circulation rate for different normal individuals. These facts suggest that there may be with some methods a constant error; and that other methods may be applicable to some normal subjects, and not to certain others.

This brings us to a consideration of certain aspects of the problem of the equilibration of lung gases with incoming venous blood, which have thus far received relatively little attention. (*a*) Some work,

as described, (6, 12, 13) has been done in comparing one method with another; the greater part of this has been in comparisons of Fick-principle with foreign-gas circulation rate methods. (b) There has been almost no work directed toward the determination of the limits of variation in composition of rebreathed mixtures, within which limits an apparent equilibrium with incoming blood can take place. And conversely, there has been a tendency to accept "plateau" or apparent equilibrium values as mixed venous values. (c) There has been relatively little study of the actual process of equilibration during the course of a single rebreathing technique.

Such studies should help to reveal the presence of false equilibrium values, or systematic differences of one method as compared with another.

In the experimental work which forms the second part of the present study, an attempt has been made to do these three things: (a) to compare three different methods, using the same subjects, these methods representing three different principles of equilibration; (b) to follow the course of equilibration by samples taken every few seconds during the procedure; and (c) (providing equilibrium levels are found) to determine the possible limits of composition of effective rebreathing mixtures.

At this point in the discussion, it may make for greater clearness if we endeavor to describe what may be expected to take place in the lungs during the course of this process of equilibration. The major variables in the process are familiar, and it should be possible to arrive at a fairly accurate and complete description.

I. Consider first the situation when "true" mixed venous equilibrium is being investigated; that is, when lung air is brought into equilibrium with both the CO_2 and the O_2 tensions of the incoming venous blood.

Take certain typical values for the various factors concerned, as shown in table 1 (many of these from an actual experiment on subject D. W. R.).

In this table it will be seen that various assumptions, as previously outlined, have been made.

A. In the first instance, suppose that, after the rebreathing of the mixture has been completed in (say) 15 seconds, (a) mixture of gases is complete, and (b) initial tensions have been such that, when mixture

is effected, O_2 and CO_2 tensions in the lung air will be the same as those of the incoming blood.

In the situation as above defined, certain points are to be noted:

1. Equilibrium will have had to be established, not only between lung air and blood, but between lung air and lung tissue fluids as well. Much of this tissue fluid is presumably in equilibrium with residual (alveolar) air. The initial mixture in the bag must, therefore, have had a certain excess of CO_2 and deficit of O_2 , in order to provide for the change of tissue fluid tension from alveolar to venous levels.

2. Equilibrium, even if it is established between lungs and blood, will not remain absolutely perfect because N_2 will tend to diffuse into

TABLE 1

Tensions of respired gases in atmospheric air, lungs, and blood with subject breathing atmospheric air

Volume of residual air in lungs = 1,700 cc.

Volume of air in rebreathing bag = 2,500 cc.

Circulation rate per minute = 4,400 cc.

Barometric pressure = 755.0 mm.

Units	Outside air (dry)	Alveolar air (= residual)	Arterial blood	Mixed venous lung tensions	Mixed venous blood tensions
Oxygen, mm.....	158.5	93.0	80.0	40.0	40.0
CO_2 , mm.....	0.2	44.6	44.6	49.8	49.8
N_2 , mm.....	596.3	569.4	569.4	617.2	569.4
Water, mm.....	0.0	48.0	48.0	48.0	48.0
Total, mm.....	755.0	755.0	742.0	755.0	707.2

venous blood, thus concentrating the lung air with respect to its other gases; therefore, a little more O_2 and CO_2 will diffuse into the blood to reestablish equilibrium. On account of the small solubility of N_2 gas in blood, this effect is negligible.

3. With the exception just mentioned, a nearly exact correspondence will be expected to exist between lung and blood gases, once equilibrium has been established, until appreciable amounts of blood begin to recirculate.

B. In the next situation to be considered, suppose the same general conditions as under A, with the further feature that a certain fraction of the blood flow has recirculated within the first 4 or 5 seconds, and

continued to do so, without any other fractions of the blood recirculating for an ensuing 10 or 15 seconds. Can a new equilibrium be obtained? In order for this to be so, it will theoretically be necessary (a) for the true venous blood to pass through the lungs unchanged (lungs at mixed venous tensions), (b) for part of this blood to recirculate through the short-circuit path (thus gaining an added increment of CO_2 and losing O_2), (c) for the new mixed venous blood (contaminated by recirculation) to make a further course through the lungs and re-establish a new venous level. Even then the new equilibrium would not be quite complete. There is no use in trying to reason too closely upon this point. It is clear that with recirculation of appreciable amounts of blood, exact plateau or equilibrium levels will not be obtained. Small amounts of blood can readily recirculate, however, without detection by present experimental methods. Thus, simple calculation will show that if there is recirculation of as much as 10 per cent of the minute volume, this will in ordinary conditions at rest, change the "equilibrium" mixed venous value by 0.5 mm. of CO_2 or less.

C. Assume that the air in the bag is such that upon mixture with residual air in the lungs, the gases are not in equilibrium with the incoming blood. The questions that arise are: (a) how nearly, at any instant, will the lung tensions be in equilibrium with the outgoing arterial (i.e., pulmonary vein) blood—i.e., is there a lag in equilibration between lungs and blood? (b) How large a discrepancy can there be at the beginning of rebreathing between mixed lung air, and venous blood, and still have enough CO_2 and O_2 absorbed into or given off from the blood within 15 to 20 seconds so as to reach equilibrium?

(a) The answer to this question could perhaps be predicted on the basis of diffusion rates of gases in pulmonary alveoli, but too little is known of these conditions to admit of even approximate estimates. The question must be decided by experiment. From an experiment which we report in detail in the subsequent part, it seems clear that for both CO_2 and O_2 in the tensions near the mixed venous values, equilibrium between lungs and blood is nearly complete at the end of the twentieth second of rebreathing.

(b) If we suppose, then, on the basis of the above mentioned experiment, that lungs and blood are in approximate equilibrium, it

becomes possible to determine from a blood nomogram the amounts of CO_2 and of O_2 that will be available from the blood, or else the amounts which the blood can be expected to absorb, when the lung gases have not the same tensions as the incoming venous blood.

Supposing the same general circulatory conditions as given above, how much O_2 and CO_2 will be available in 15 or 20 seconds if the mixed lung gases (CO_2 and O_2) at the outset are 5 mm. lower than the venous gases; and how nearly will equilibrium be reached within this time? Conditions at the start will be as in table 2 (column 3 here is the same as column 6 of table 1, column 2 gives the assumed initial lung tensions).

Presumably, equilibrium between lung air and incoming blood in the alveoli will be approached asymptotically. We have not attempted

TABLE 2
Tensions of respired gases in low O_2 - CO_2 rebreathing mixture compared with tensions in normal venous blood

	Initial mixed lung tensions	Mixed venous blood tensions
Oxygen, mm.	35.0	40.0
CO_2 , mm.	44.8	49.8
N_2 , mm.	627.2	569.4
Water, mm.	48.0	48.0
Total, mm.	755.0	707.2

an integral expression for this change, but have assumed the pressure gradient to remain constant over 1 second intervals and in the case of the CO_2 , have calculated the CO_2 tension in the lungs at each particular second by adding to the initial amount of CO_2 in the lungs, the sum of the amounts of this gas contributed from the blood during each of the preceding seconds, then dividing this number by the total volume of lung-bag air and multiplying by the barometric pressure. The difference between the lung tension thus calculated and the mixed venous CO_2 tensions then gives the pressure gradient for that particular second of time.

By this calculation, it is found that at the end of 15 seconds, the CO_2 tension in the lung-bag system will be 47.8 mm., still 2 mm. below the

mixed venous level, and at the end of 20 seconds 48.4 mm., or 1.4 mm. below equilibrium.

For oxygen, the values at the end of 15 and 20 seconds will be even further from equilibrium because the dissociation curve is flatter than the CO₂ dissociation curve, that is, a smaller volume of oxygen is given off from the blood per millimeter of pressure difference, than of CO₂.

It may be added here parenthetically that a simpler and sufficiently

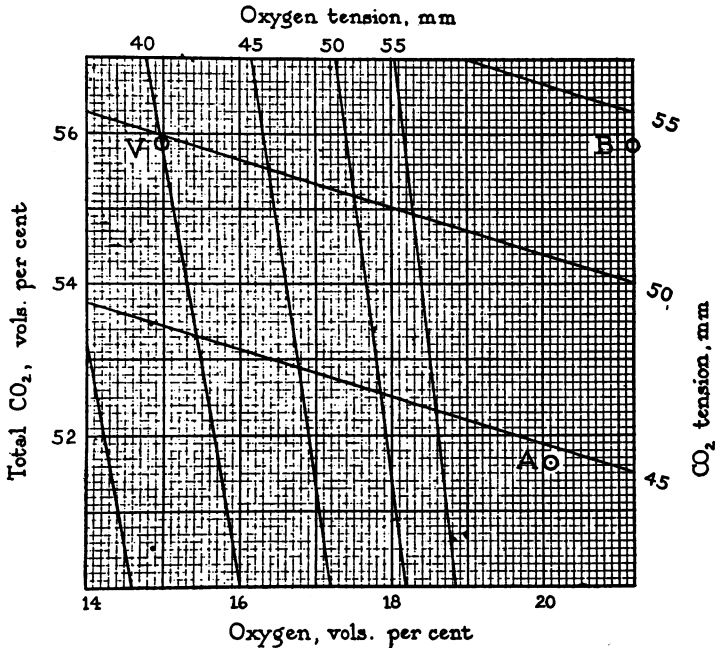


FIG. 1. CO₂-O₂ NOMOGRAM OF SUBJECT D. W. R.

A, arterial blood; V, mixed venous blood; B, oxygenated mixed venous blood

accurate method for making this calculation is by assuming over the whole 20 seconds of rebreathing a constant average pressure gradient, namely half the initial pressure difference between lung-bag air and blood, or 2.5 mm. for both CO₂ and O₂. In this region of the nomogram (fig. 1), 1 mm. change of CO₂ tension at constant oxygen tension corresponds to 0.51 cc. CO₂ (under standard temperature and barometer conditions); and 1 mm. of change of O₂ tension at constant

CO₂ tension, to 0.23 cc. O₂. In 20 seconds, there will be $2.5 \times 0.51 \times 20 \times \frac{4400}{60 \times 100} \times \frac{273 + 38}{273} = 21.4$ cc. of CO₂ given off by the blood (at 38°C.) and $2.5 \times 0.23 \times 20 \times \frac{4400}{60 \times 100} \times \frac{273 + 38}{273} = 9.6$ cc. of O₂. The 21.4 cc. of CO₂ would raise the CO₂ tension from 44.8 to $\left\{ \left[\left(\frac{44.8}{755} \times 4200 \right) + 21.4 \right] \div 4231 \right\} \times 755 = 48.4$ mm.; and the 9.6 cc. of O₂ would raise its tension from 35 mm. to $\left\{ \left[\left(\frac{35}{755} \times 4200 \right) + 9.6 \right] \div 4231 \right\} \times 755 = 36.6$ mm.

These estimates are naturally only first approximations, but it is clear that, in order for equilibrium to be obtained by this method between lungs and blood within 20 seconds, the differences in tension between the gas tensions of lungs and blood, at the start of equilibration must be small; less than 5 mm. for CO₂ and considerably less than this for O₂. In the experiments reported in the subsequent part, this conclusion will be verified.

II. In the case where oxygenated mixed venous blood is being studied, a further set of typical values will be useful (table 3).

A. Again we assume here that mixture is complete, and, in the first instance, that the tensions in lungs and bag are such that when mixture has been effected, and incoming blood has been completely oxygenated, then the CO₂ tension in lung air and incoming blood will be the same. In other words, there will be no net change of the CO₂ of the blood during its passage through the lungs. During the first part of the length of the alveolar capillary, while the blood is still partly unsaturated with O₂, there will be a small temporary inflow of CO₂, which will leave the blood again as arterialization is completed. This transient process is here disregarded.

In this instance, however, it becomes necessary to investigate the effect (already referred to above), of O₂ absorption by the blood, and consequent concentration of other lung gases, upon the general equilibrium values.

By reference to the CO₂-O₂ nomogram, figure 1, it is found that the normal venous blood of D. W. R. with an O₂ tension of 40 mm.,

and CO₂ tension of 50 mm., contains 15.0 cc. of O₂ per 100 cc. of blood. The oxygenated blood leaving the lungs after rebreathing the high O₂ mixture will be (it is assumed) completely saturated (Field, Bock, et al. (16)), and will in this case contain 21.2 cc. of O₂ per 100 cc. of blood. About 6.2 cc. of O₂ will, therefore, be absorbed per 100 cc. of blood flowing. In 20 seconds, with a minute volume of circulation of 4400 cc. this will amount to $44 \times \frac{20}{60} \times 6.2 = 91$ cc. of O₂ absorbed. This latter figure represents O₂ under standard temperature and bar-

TABLE 3

Comparison of tensions of respired gases in (a) atmospheric (outside) air, (b) alveolar air, (c) arterial blood, (d) and (e) lungs and blood in "oxygenated mixed venous" equilibrium

Volume of residual air in lungs = 1,700 cc.
 Volume of air in rebreathing bag = 2,500 cc.
 Circulation rate per minute = 4,400 cc.
 Barometric pressure = 755 mm.

	Outside air (dry)	Alveolar air (= residual)	Arterial blood	Oxygenated mixed venous lung tensions	Oxygenated mixed venous blood tensions
Oxygen, mm.....	158.5	93.0	80.0	100+	100±
CO ₂ , mm.....	0.2	44.6	44.6	54.0	54.0
N ₂ , mm.....	596.3	569.4	569.4	553-	553±
Water, mm.....	0.0	48.0	48.0	48.0	48.0
Total, mm.....	755.0	755.0	742.0	755.0	755(?)

ometric conditions: $91 \text{ cc.} \times \frac{273 + 38}{273} = 104$ cc. of lung-bag oxygen absorbed. This will change the CO₂ tension in lung-bag system from its initial value of $\frac{300}{4200} \times 755 = 54.0$ mm. to $\frac{300}{4096} \times 755 = 55.3$ mm.

The ultimate effect will be less than this, because the extra CO₂ will redistribute itself between lungs and blood, and much of it, therefore, will be carried away in the blood as fast as it accumulates. As an approximation, we may suppose an average CO₂ tension gradient between lungs and blood (due to oxygen absorption), of $\frac{55.3 - 54.0}{2} =$

0.65 mm. Then in 20 seconds, $0.65 \times 0.51 \times \frac{4400}{300} = 4.9$ cc. of CO_2 reabsorbed. Thus the final lung-bag tension of CO_2 at the end of 20 seconds will be approximately $\frac{300 - 4.9}{4096} \times 755 = 54.4$ mm.

In summary, therefore, this concentration effect is not quite negligible; it should cause the oxygenated mixed venous tensions of CO_2 , at the end of 20 seconds rebreathing, to be in the region of 0.4 mm. higher than the true equilibrium value; and it should further provide that, even if this true equilibrium value is attained at any moment, the CO_2 tension will thereafter show a small steady rise, instead of remaining perfectly constant. This latter effect would be small, however, and by itself would probably escape detection with present methods of analysis.

If, as suggested above, the oxygenated mixed venous blood is actually not more than 97 per cent saturated, it can readily be seen that the error involved in assuming complete oxygenation will tend to counteract the error due to neglecting the oxygen absorption effect.

B. Assume in this instance that lung-bag air (after mixture) is such that upon oxygenation of the blood, the tension of the CO_2 in the blood thus oxygenated is *less* than the CO_2 tension in lung-bag air. CO_2 will pass from lungs into blood. The question arises, how much of an excess of CO_2 can exist in lung-bag air, and still reach an equilibrium with the incoming venous blood at the end of 20 seconds? How much extra CO_2 can the blood absorb?

The attempt to calculate how much CO_2 will be absorbed into the blood, with a given CO_2 tension difference between blood and lung-bag air, raises the further question as to how complete is the equilibrium between lungs and (oxygenated) alveolar blood, with respect to CO_2 . In the region of the tensions of oxygenated mixed venous blood this equilibrium has been shown by Field, Bock, et al. (16), to be nearly complete; rebreathed air samples having been compared with simultaneously drawn arterial blood. It is, therefore, again possible to determine from a blood nomogram, or from a CO_2 dissociation curve of arterial blood, approximately the amounts of CO_2 that can be absorbed in a given interval.

In this calculation account must be taken of the oxygen absorbed

from lung-bag air during the interval. Taking the same basic figures, we have 104 cc. of O_2 absorbed in 20 seconds, out of a total of lung-bag volume of 4200 cc.

Suppose that the lung-bag mixture is 2 mm. higher in CO_2 tension than that of the oxygenated mixed venous blood. Take the same figures as before for volumes, blood gas tensions, blood flow, etc.

At the start of rebreathing, the lungs will contain $\frac{56}{755} \times 4200 = 312$ cc. of CO_2 . In 20 seconds $\frac{4400}{3}$ cc. of blood will flow. For every 1 mm. change in CO_2 tension, 0.46 cc. of CO_2 will be absorbed per 100 cc. blood (see nomogram, from 50 mm. to 55 mm. CO_2 tension on completely oxygenated line). With an average pressure head of 1.5 mm. the CO_2 absorbed in 20 seconds will be $\frac{4400}{300} \times 0.46 \times 1.5 = 10.1$ cc. $312 - 10.1 = 302$ cc. of CO_2 remaining in lung-bag air. The volume of this air will now be $4200 - 104 - 10.1 = 4086$ because of loss of O_2 and CO_2 . The final CO_2 tension will, therefore, be $\frac{302}{4086} \times 755 = 55.8$ mm. In other words, even with the small excess of 2 mm. of lung-bag CO_2 tension, equilibrium cannot theoretically be reached with the incoming venous blood, within 20 seconds. This calculation is admittedly approximate and several small corrections that might be made have been omitted, but the tentative conclusion that it leads to seems unmistakable.

A similar calculation for an excess of 5 mm. in lung-bag mixture (initial CO_2 tension of 59 mm.) leads to a final CO_2 tension (at the end of 20 seconds) of 57 mm.; and an initial excess of 10 mm. of CO_2 (64 mm. tension) gives a tension after 20 seconds of 58.5 mm. Thus it is evident that with increasing excesses of CO_2 in the lung-bag air, the final values at the end of rebreathing steadily, though quite gradually, rise above the tensions of CO_2 of oxygenated mixed venous blood. This type of change will be apparent in the experimental data given later.

It should be noted that in the preceding comments no account has been taken of the buffering effect, so to speak, of the lung tissues themselves. These tissues will, for the most part, presumably be near the gaseous tensions of the alveolar air, and will, therefore, absorb certain

amounts of CO₂ when equilibrium with mixed venous tensions is being approached. It would be difficult to assign any values to the amounts of CO₂ thus absorbed.

C. Finally, it is necessary to consider the case in which the lung-bag mixture is lower in initial CO₂ tension than the oxygenated mixed venous blood. As pointed out above, most of the methods used hitherto, and all of those based on the Henderson-Prince technique, are in this class, the equilibrium values for CO₂ tension being approached from below. What is the lowest initial CO₂ tension that may be used and still establish equilibrium with the incoming venous blood before recirculation?

Again let us take the same physiological values as before. Suppose the lung-bag mixture to be 4 mm. lower in CO₂ tension than the oxygenated mixed venous tension, that is, 50 mm. At the start, in the lung-bag space there will be $\frac{50}{755} \times 4200 = 278$ cc. CO₂. In 20 seconds approximately $\frac{54 - 50}{2} \times \frac{4400}{300} \times 0.46 = 13.5$ cc. of CO₂ (standard conditions) will be given off from the blood; this at lung temperature will be about $\frac{311}{273} \times 13.5 = 15.4$ cc. The CO₂ tension in the lung-bag mixture after rebreathing will, therefore, be $\frac{278 + 15.4}{4200 - 104 + 15.4} \times 755 = 53.9$ mm. This is practically the actual oxygenated mixed venous CO₂ tension.

If the initial lung-bag mixture were 6 mm. too low in CO₂ tension, then the value after rebreathing would be 53.2 mm., or 0.8 mm. below equilibrium value.

Again no account has been taken of the effect of the lung tissues, which will require certain extra amounts of CO₂ to raise them to oxygenated mixed venous levels.

Thus, under the conditions here postulated, the lowest possible value of initial lung-bag CO₂ tension, that can be equilibrated with the venous blood within 20 seconds will be a tension somewhat less than 4 mm. below the oxygenated mixed venous tension.

Now apply this same type of calculation to an experimental method

in which successive rebreathing periods are employed. Supposing that following a certain period of rebreathing, an equilibrium with the oxygenated mixed venous blood is reached, with respect to CO₂; in the next period will the dilution due to residual alveolar air be made up by CO₂ given off from the blood?

Take for the purpose a series of actual data from subject D. W. R.:

Alveolar CO₂ tension 43.9 mm.

Oxygenated mixed venous CO₂ tension 53.3 mm.

Residual air 2,000 cc. containing $\frac{43.9}{755} \times 2,000 = 116$ cc. CO₂.

Air in bag 2,500 cc. containing $\frac{53.3}{755} \times 2,500 = 176$ cc. CO₂.

Lung-bag tension at start of rebreathing $\frac{116 + 176}{2000 + 2500} \times 755 = 49.0$ mm. CO₂.

Circulation rate per minute = 5100 cc. per minute.

CO₂ available from blood (see above):

$$\frac{53.3 - 49.0}{2} \times \frac{1700}{100} \times 0.46 \times \frac{311}{273} = 19.2 \text{ cc. in 20 seconds.}$$

$$14.4 \text{ cc. in 15 seconds.}$$

Total air in lung-bag system after 20 seconds:

$$4500 + 19.2 - 120 = 4399 \text{ cc. (120 cc. of O}_2 \text{ absorbed in 20 seconds).}$$

$$\text{After 15 seconds, } 4500 + 14.4 - 90 = 4424.$$

Final CO₂ tension after rebreathing:

$$\frac{292 + 19.2}{4399} \times 755 = 53.4 \text{ mm. at end of 20 seconds.}$$

$$\frac{292 + 14.4}{4424} \times 755 = 52.4 \text{ mm. at end of 15 seconds.}$$

In this case, then, there is just enough CO₂ available to return the tension to its former value after 20 seconds' rebreathing, providing no account is taken of absorption of CO₂ by lung tissue fluids. If this is appreciable, then the values of CO₂ reached by successive rebreathings for 20 seconds will not represent the actual mixed venous level. Still further deviations will occur if (a) the residual air is greater in volume relative to the air in the bag, (b) blood flow is slower, and (c) arterio-venous difference is greater. It will, therefore, be clear that for a technique of this sort, the actual mixed venous value may be obtained with one subject and not obtained with another.

SUMMARY

Preliminary to some experimental studies upon the mixed venous blood values of man at rest, the previous literature has been reviewed, with special emphasis upon the inferential nature of the results obtained, and the numerous assumptions involved.

Certain aspects of the process of equilibration of lung gases with incoming venous blood, are discussed.

PART II

EXPERIMENTAL

In the first part of the present investigation we have reviewed the previous work on this subject, discussed some of the assumptions upon which the measurements of mixed venous blood in man are based, and described what we conceived to be the probable sequence of events within the lungs when the rebreathing procedures were carried out under various conditions.

In our experimental work, we have attempted to study further the process whereby equilibrium between lung air and incoming venous blood is reached; specifically to find answers to the following questions:

1. Is a "plateau" or apparent equilibrium level in rebreathed lung gases demonstrable over an appreciable length of time, before evidence of recirculating blood is manifest?

2. If such an apparent equilibrium between respired gases and incoming blood is reached, how soon after beginning rebreathing can this be demonstrated, how long does it last, how soon do signs of recirculation of blood appear and how rapidly does this recirculating blood change the apparent equilibrium values?

3. What are the upper and lower limits of respired air mixtures, within which limits an apparent equilibrium can be reached before recirculation of blood occurs?

4. Are oxygenated mixed venous values comparable with actual or true mixed venous values?

5. Do the methods of intermittent rebreathings (without changing the gas in the bag between periods) give the same values as methods using only a single rebreathing period?

6. Are lungs and arterial blood in equilibrium during the rebreathing procedures (only a partial answer attempted)?

The experimental studies which we have made fall under three general headings:

(a) The equilibrium in the case of oxygenated mixed venous tensions, during a single rebreathing period.

(b) The same, studied during intermittent rebreathing periods, with intervals of rest between, and without changing the gases in the rebreathing bag in these intervals.

(c) The equilibrium in the case of the actual mixed venous tensions (equilibrated both with CO_2 and O_2), during a single rebreathing period. Each of these determinations was always either preceded or followed by an equilibration with oxygenated tensions (group (a) above), for purposes of comparison.

Two normal subjects were used, M. L. S. and D. W. R., a considerable number of experiments having been performed with the latter, and a smaller number—with the major purpose of confirming the general findings on D. W. R.—with the former subject. A few relevant data regarding the two subjects are listed in tables 4 and 5.

It is evident that for the purposes of this study it was important to have as many observations as possible taken during the course of a single rebreathing period; this was done by the use of a series of evacuated gas sampling tubes, either six or nine being used to give the corresponding number of separate samples during the experimental period of from 25 to 50 seconds.

METHODS

In further detail, the methods used were as follows:

(a) For oxygenated mixed venous tensions, the simple apparatus shown in figure 2 was employed. Most of the experiments were done toward the end of the morning, the subject having had an ordinary breakfast three or four hours before. A few were done in mid-afternoon, two or three hours after lunch. After 15 minutes to half an hour of rest in the horizontal position, the subject applied mouth-piece and noseclip and respired through the apparatus for five to ten minutes. At the end of this time, at the signal "Blow," given at the end of a normal expiration, the subject expired completely while slide valve B was closed off. At the end of this expiration an alveolar air sample was (usually) taken into one of the sampling tubes. After a rest interval of three minutes, a similar complete expiration was

TABLE 4

Subject D. W. R., male, aged thirty-three, height, 188 cm.; weight, 75 kgm.; vital capacity, 6300 cc.; residual air, 2000 cc.

Date	Initial P _{CO₂} in bag		Alveolar P _{CO₂}		Oxygenated mixed venous CO ₂ tensions											
	mm.	mm.	mm.	mm.	5-second sample	10-second sample	15-second sample	20-second sample	25-second sample	30-second sample	35-second sample	40-second sample	45-second sample	50-second sample	Δ	
Group I (50 to 56 mm.):																
November 20.....	52.5	48.6	+2.3	50.9	+1.3	52.2	+1.1	53.3	+0.9	54.2	+0.7	54.9	(55.0)			
December 3.....	55.8	52.5		53.3	+1.4	54.7	+0.7	55.4	(55.0)							
December 3.....	55.1	51.7		53.0		54.1	+1.1	55.2	+0.3	55.5		55.9	(55.6)			
December 7.....	55.9	52.7	0.0	52.7		54.1	+1.1	55.7	+0.2	55.9	(55.5)					
December 7.....	56.0	51.8	+0.1	51.9		53.9	+1.2	55.1	-0.4	54.7	(54.7)					
December 17, a.m.....	53.6	51.8		53.4		54.6	+0.3	54.9	+1.0	55.9	(55.3)					
December 17, p.m.....	54.0	52.3	+0.5	52.8		54.8	+0.4	55.2	+1.0	56.2	(55.9)					
December 18, a.m.....	51.0	44.1		51.1		53.8	+1.0	54.8	+0.4	55.2	+1.0	56.2	(55.9)			
December 21, a.m.....	54.2	43.0		51.3		53.0	+0.4	53.0	+0.5	53.5	(53.4)					
December 26, a.m.....	51.7	45.8		52.2		51.9	+0.5	52.4	+0.6	53.0	+0.8	53.8	(53.0)			
February 18, a.m.....	52.5	41.3		49.6		53.8	+0.9	54.7	+0.5	55.2	(54.8)					
Average.....	53.8	43.6		51.7		53.7	+0.7	54.4	+0.5	55.0		55.9				
Group II (56 to 60 mm.):																
December 10.....	56.7	53.0	+0.6	53.6		55.1	+0.7	55.8	(55.2)							
December 10.....	56.6	52.3	+0.6	52.9		54.2	+0.5	54.7	+0.6	55.3	(55.0)					

TABLE 4—Concluded

Date	Initial P _{CO₂} in bag	Alveolar P _{CO₂}	Oxygenated mixed venous CO ₂ tensions																	
			5-second sample	10-second sample	15-second sample	20-second sample	25-second sample	30-second sample	35-second sample	40-second sample	45-second sample	50-second sample								
Group IV (65 to 70 mm.)—Concluded:																				
December 31, p.m....	68.5	46.8	54.1	+0.7	54.8	0.0	54.8	0.0	53.6	+0.4	54.0	+0.6	55.9	+0.4	56.3	+0.1	56.4	+1.4	57.8	(57.7)
February 16, a.m....	67.2	43.2	53.2	-0.4	52.8	+0.8	53.6	0.0	53.6	+0.4	54.0	+0.4	54.1							
March 15, a.m....	66.4	44.8	52.7	-0.1	52.6	+0.1	52.7	+0.4	53.1	0.0	53.1	0.0	53.3							
Average.....	67.2	44.4	53.3	-0.1	53.3	+0.1	53.3	+0.1	53.1	+0.4	53.8	+0.3	54.2	+0.4	56.3	+0.1	56.4	+1.4	57.8	
Group V (70 to 80 mm.):																				
November 20, a.m....	77.8		59.8	-0.5	59.3	-1.1	58.2	-0.6	57.6	-0.6	57.0	+0.4	57.4	(57.9)						
December 21.....	72.0	43.7	54.8		53.7		54.0	+0.6	54.6	-0.4	54.2	(54.8)								
January 2, p.m....	73.4	44.2	56.0	-0.1	55.9	+0.3	56.2	+0.1	56.3	+0.6	56.9									
January 3, a.m....	76.4		57.4	-1.1	56.3	0.0	56.3	+0.2	56.5	+0.4	56.9	+0.1	57.0	(57.1)						
January 3, a.m....	79.2		57.2	-1.0	56.2	-0.4	55.8													
January 5, a.m....	79.8		59.0	-1.6	57.4	-0.9	56.5	+0.4	56.9	+0.1	57.0	-0.4	56.6	(57.5)						
March 7, a.m....	76.2	42.2	54.6	-0.9	53.7	-0.4	53.3	+0.2	53.5	0.0	53.5	(53.3)								
May 13, a.m....	71.1	44.5	55.2	0.0	55.2	+0.2	55.4	+0.4	55.8	-0.1	55.7	(56.0)								
Average.....	75.7	43.7	59.8	-0.8	56.7	-0.8	56.1	-0.2	55.9	+0.1	55.7	+0.3	55.9	0.0	56.0	+0.9	58.6			

TABLE 5
 Subject M. L. S., female, aged twenty-nine, height, 170 cm.; weight, 58 kgm.; vital capacity, 4000 cc.

Date and time	Initial P _{CO2} in bag	Alveolar P _{CO2}	Oxygenated mixed venous CO ₂ tensions												
			10-sec-ond sample	Δ*	15-sec-ond sample	Δ	20-sec-ond sample	Δ	25-sec-ond sample	Δ	30-sec-ond sample	Δ	35-sec-ond sample	Δ	
	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.
Group I (45 to 50 mm.):															
March 2, a.m.	45.9	36.8	43.8	+0.5	44.3	+0.6	44.9	+0.6	44.9	+0.6	45.5	+0.6	46.1	+0.6	(45.5)
March 2, a.m.	48.9	37.4	44.9	+0.5	45.4	+0.5	45.9	+0.2	46.1	+0.6	46.7	+0.6	46.7	+0.6	(46.4)
Average	47.4	37.1	44.4	+0.5	44.9	+0.6	45.4	+0.4	45.8	+0.6	46.4	+0.6	46.4	+0.6	
Group II (50 to 55 mm.):															
January 18, a.m.	50.4	39.7	47.3	0.0	47.3	0.0	47.3	+0.4	47.3	+0.4	47.7	+0.1	47.8	+0.1	(47.8)
January 18, a.m.	50.3	39.7	46.5	-0.5	46.0	+0.4	46.4	+1.1	47.5	+0.1	47.6	+0.1	47.6	+0.6	48.2
January 21, a.m.	50.2	37.3	46.4	+0.8	47.2	0.0	47.2	-0.4	46.8	+0.4	47.2	+0.4	47.2	+0.6	(48.3)
January 21, p.m.	51.8	40.0	47.9	+0.3	48.2	+0.7	48.9	+0.3	49.2	+0.6	49.8	+0.6	49.8	+0.6	(49.6)
February 25, a.m.	52.4	36.7	45.1	+0.1	45.2	0.0	45.2	+0.3	45.5	+0.5	46.0	+0.5	46.0	+0.5	(45.9)
March 1, a.m.	51.7	37.2	46.4	-0.1	46.3	+0.9	47.2	0.0	47.2	+0.5	47.7	+0.5	47.7	+0.5	(48.0)
Average	51.1	38.2	46.6	+0.1	46.7	+0.3	47.0	+0.3	47.3	+0.4	47.7	+0.4	47.7	+0.6	48.2
Group III (55 to 65 mm.):															
January 21, p.m.	59.8	39.6	47.7	+1.8	49.5	-0.1	49.4	+0.4	49.8	+0.2	50.0	+0.2	50.0	+0.2	(49.7)
January 21, a.m.	60.1	38.5	47.9	+1.1	49.0	0.0	49.0	+0.2	49.2	-0.1	49.1	-0.1	49.1	-0.1	(48.7)
Average	60.0	39.1	47.8	+1.5	49.3	-0.1	49.2	+0.3	49.5	+0.1	49.6	+0.1	49.6	+0.1	

* Columns headed Δ show change between successive 5-second samples.

made at the "Blow" signal, the slide valve shut down as before, and the 3 way valve turned so as to connect the rebreathing bag with the subject. The subject then emptied and filled the bag completely, by successive respirations, samples being taken into the sampling tubes every 5 seconds at the end of a complete expiration. At first a complete respiration was made every $2\frac{1}{2}$ seconds (subject timing his breathing with the aid of a stop watch, starting the watch at the "Blow" signal). The ten second sample was taken after the third complete respiration, the 15 second sample after the fifth, etc. It was later found that as rapid and complete a mixture of rebreathed gases could apparently be obtained by respiring only once in 5 seconds, holding the air in the lungs for about 2 seconds following each inspiration. At the end of the expiration when the last sample was taken, valve E was turned again, thus shutting off the air in the bag, and the experiment was ended.

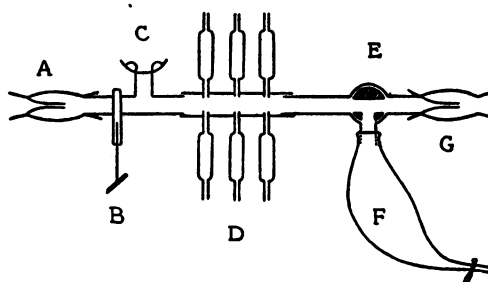


FIG. 2. DIAGRAM OF APPARATUS USED FOR OXYGENATED MIXED VENOUS TENSION DETERMINATIONS

A, intake flutter valve; B, shut-off slide valve; C, mouthpiece; D, six evacuated gas sampling tubes; E, three-way valve; F, rebreathing bag, containing $\text{CO}_2\text{-O}_2$ mixture; G, outgo flutter valve.

A sample of the bag air was taken, immediately, to compare with that in the last evacuated sampling tube.

Usually two such experiments were done in succession, the subject remaining at rest in the 10 to 15 minute interval between experiments. In some of the earlier experiments the subject did not remain at rest, but got up and moved about doing ordinary laboratory work.

(b) In the case of the experiments done with successive rebreathing periods the apparatus was the same, but the technique of collection differed slightly; at the end of a normal respiration, as before, the signal was given, the subject expired completely, slide valve B was closed, and an alveolar sample taken. After three minutes rest, breathing through the apparatus, the same technique was repeated, except that after slide valve B was closed, valve E was turned to connect bag and subject. The subject then rebreathed the gas mixture four times in about 18

seconds. At the end of the fourth expiration, one sample was taken into an evacuated tube, valve E turned again to connect patient with outgo valve, and shut-off valve B opened. The subject rested another 3 to 5 minutes (mouthpiece and noseclip still in place). Then another similar rebreathing was done, as described. This process was repeated every 3 to 5 minutes for five to six periods.

Following this experiment, after a further 10 to 15 minute rest, a single-period experiment (as in (a) above) was carried out, for comparison.

Samples were analyzed for CO_2 with the Haldane apparatus.

The rebreathing bag contained 2.5 liters, the mixture consisting of oxygen plus varying amounts of CO_2 . In order to determine the limits for initial CO_2 tension

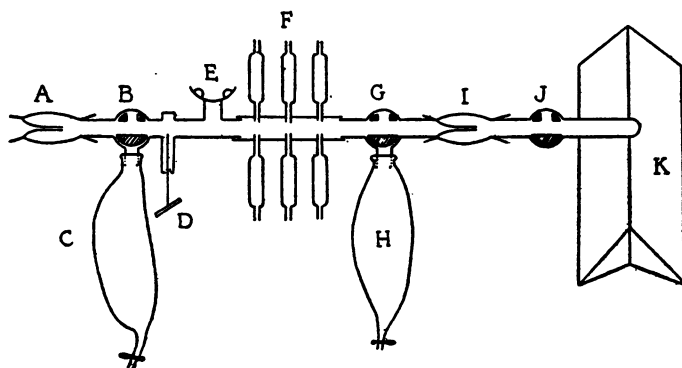


FIG. 3. DIAGRAM OF APPARATUS USED FOR TRUE MIXED VENOUS TENSION DETERMINATIONS

A, intake flutter valve; B, three-way valve; C, rebreathing bag, containing preliminary CO_2 - N_2 mixture; D, shut-off slide valve; E, mouthpiece; F, six evacuated gas sampling tubes; G, three-way valve; H, rebreathing bag, containing CO_2 - N_2 - O_2 mixture; I, outgo flutter valve; J, three-way valve; K, Douglas bag for collection of expired air.

in the bag, within which limits an apparent equilibrium could be obtained by rebreathing, widely different initial tensions were employed, varying from 45 mm. to 80 mm., as shown in tables 4 and 5.

(c) For the determination of actual mixed venous tensions, a slightly modified apparatus was used, based on the technique described by Burwell and Robinson (20). A diagram of the apparatus is given in figure 3.

Bag C was filled with nitrogen plus a small amount—from 30 mm. to 50 mm.—of CO_2 ; this mixture was rebreathed twice, for the purpose of lowering the O_2 tension in the residual air. Bag H was filled with a mixture containing about 55 mm. CO_2 , 35 mm. O_2 , and the rest nitrogen. This latter mixture was worked out, partly by calculation and partly by trial and error, so as to make the resultant gas

tensions, after mixture by rebreathing, as nearly as possible the same as those of the incoming venous blood; i.e., so as to reduce the necessary equilibration to a minimum.

The technique of these experiments, in further detail, was as follows: After the preliminary rest period, and the five to ten minutes of breathing through the apparatus, the signal "Blow" was given, as before, at the end of a normal expiration, and a complete expiration made. Valve B was then opened from bag C to the subject, and valve J closed off. Two complete respirations were made into bag C. Slide valve D was then closed and valve G opened between bag H and the subject. Complete respirations were then made every 5 seconds, a sample being taken into one of the sampling tubes at the end of each, and the exact second of taking noted. The stopwatch was started at the signal "Blow." When the last sample was taken, at the end of about thirty seconds, valve G was closed to the bag and the experiment ended.

The subject was usually moderately cyanosed and dyspneic following these experiments.

After a fifteen to twenty minute rest period, a set of oxygenated mixed venous tensions was obtained as described under (a) above. On one occasion the order was reversed, the oxygenated experiment being done first. No difference was noted.

In certain experiments, for the purpose of obtaining the respiratory quotient, a 5 or 6 minute volume of expired air was collected in a Douglas bag, before the re-breathing was begun, and analyzed for O_2 and CO_2 . A sample of room air was usually also analyzed, as the intake valve did not lead in from the outside air.

RESULTS

(a) *Oxygenated mixed venous blood tensions*

In tables 4 and 5 are given the results, according to the technique described above, of forty-six consecutive experiments with subject D. W. R., and of ten experiments with subject M. L. S. In the case of the former subject this represents every experiment that was done except for two preliminary ones with a slightly different technique. In the case of M. L. S., one day's experiments were discarded because of leaks discovered in the sampling tubes. Every CO_2 determination that was made is also given, but certain values which are enclosed in brackets, are not included in the further calculations, as being improbably far from other values.

In the tables, it will be noted that the second column gives the CO_2 tensions in the bag before rebreathing was begun, and the next column the CO_2 tension of the alveolar air. The fourth, sixth, etc. columns

give the CO₂ tensions in the fifth-, tenth-, fifteenth-, etc., second samples, and the intervening columns (fifth, seventh, etc.) give the CO₂ tension differences between these successive 5-second values. The figures in parentheses are the CO₂ tensions of samples taken from the rebreathing bag after the procedure was ended; these in general are close to the values of the last of the samples taken in the evacuated sampling tubes.

The experiments are listed in five groups, according to the initial CO₂ tensions in the rebreathing bag; Group I, 50 to 56 mm.; Group II, 56 to 60 mm.; Group III, 60 to 65 mm.; Group IV, 65 to 70 mm.; Group V, 70 to 80 mm. The separation of Groups I and II at 56 mm. is made because of the well defined difference that occurs in the series below this value as compared with the series above.

Averages of all figures are given at the end of each group.

In figure 4, these averages for subject D. W. R. are presented in graphic form. It will be noted that there is in this figure an ordinate marked "Estimated mixed CO₂ tension." This is an estimate of the concentration of CO₂ in the mixed respired gases, calculated from the volume of bag air, its CO₂ content, the volume of residual air,¹ and its CO₂ content. It is assumed that no absorption or elimination of gases by the blood has taken place. The resultant figure gives an approximate measure of the extent of change required for the establishment of equilibrium between lungs and incoming venous blood.

Considering for the moment only the experiments on D. W. R., one finds certain points brought out:

1. In the group where the initial bag tensions were 50 to 56 mm. it is clear that no consistent "plateau" level is reached at any time. There is thus no evidence that an equilibrium has been achieved between lung gases and incoming venous blood. There are considerable differences of tension between successive sets of samples taken at

¹ The subject's residual air was calculated by having him expire completely, then rebreathe from a spirometer a high O₂ mixture several times in 15 seconds, (to obtain complete mixing), then again expire completely into the spirometer. The nitrogen in residual air was assumed to be 75 per cent of the total gas volume (25 per cent taken by CO₂, O₂, and water vapor). From this value, the volume of the gases in the spirometer at the end of rebreathing, and the percentages of N₂ in the spirometer at the beginning and end of rebreathing, the approximate value of the residual air can readily be calculated. For subject D. W. R. this volume was about 2000 cc.

5 second intervals. (On the chart there is an apparent equilibrium from 10 to 15 seconds, but reference to the table will show that this is false, and due to the fact that for certain experiments no 10-second sample was obtained.)

2. In the next three groups, with initial tensions at 56 to 60 mm., 60 to 65 mm., and 65 to 70 mm., there is a clear cut "plateau" found

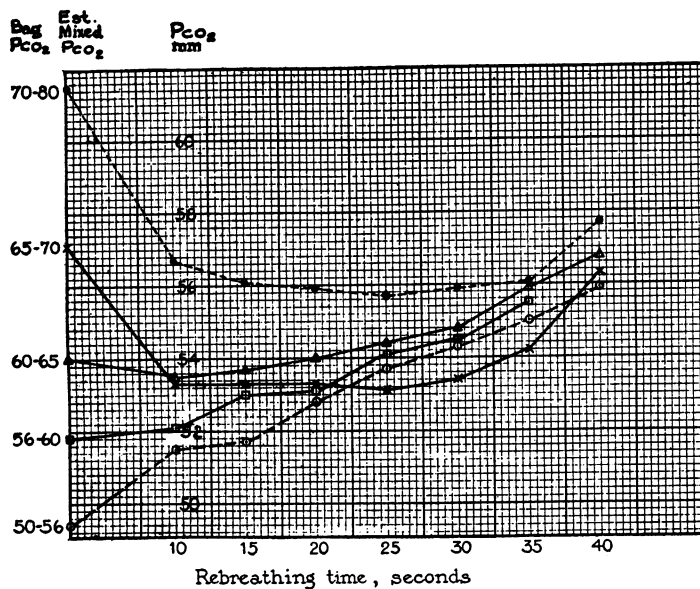


FIG. 4. OXYGENATED MIXED VENOUS CO₂ TENSIONS

Ordinates, tensions of CO₂, mm. (= P_{CO₂}). Abscissae, time at which sample was taken, expressed as the number of seconds after rebreathing was begun. The "Bag P_{CO₂}" figures indicate the initial tensions in the rebreathing bag, for the particular group of experiments represented by the adjacent line and symbol. For explanation of "Estimated P_{CO₂}" tensions, see text. Each line represents the average CO₂ tension change during rebreathing, for that group of experiments.

between the fifteenth and twentieth second. This is demonstrated not only in the average figures, but in almost every one of the individual experiments: thus, of the twenty-seven experiments, there were two differences of +0.8 mm. CO₂ in the 15 to 20 second interval, one of +0.6 mm., and one of +0.4 mm., all the rest being between

± 0.3 mm. As the error of method in the Haldane gas analyses is itself about 0.2 mm., this degree of variation can be considered small. Thus there is found at the plateau level both a decrease in the amount of average change in level over the interval, as compared with earlier and later intervals, and a decrease in the deviations of individual values from the average, when compared with similar deviations for the differences in other time intervals.

Even over this plateau, however, it is clear that there is a small rise in CO_2 tension in most instances, with an average change of $+0.19$ mm. CO_2 . This may be due to the recirculation of increasing small amounts of blood, or it may be in part, at least, an effect due to oxygen absorption, discussed in the previous part. In either case, the plateau level should then be slightly above the actual oxygenated mixed venous level.

3. Beginning at 25 seconds, there is a progressive rise in CO_2 level, presumably due to increasing amounts of recirculating blood.

4. The average CO_2 tension level in the three groups is nearly the same, being (at the fifteenth second) 53.0 mm. for Group II, 53.7 mm. for Group III, and 53.3 mm. for Group IV. The fact that with these considerable variations in initial bag mixture, the plateau level is nearly the same is further evidence that this plateau represents a true equilibrium condition.

5. Although average plateau levels are nearly the same, individual experiments show considerable variations in this respect. The lowest figure at 15 seconds is 52.6 mm., and the highest 55.9 mm. Such variations are to be expected. The work of Higgins (29) and later of Dodds (30) on alveolar air values, during the course of the day, and that of Cullen and Earle (31) more recently, on the arterial pH_a and CO_2 tensions, show that considerable variations take place, and it is reasonable to suppose similar changes in mixed venous levels. There are, in fact, some early experiments by Porges, Leimdörfer, and Markovici (32), using the original Plesch method, which showed such variations in mixed venous tensions. It is clear, for instance, from our data, that the mixed venous tension level is usually higher in the afternoons than in the late mornings. A further point, which is obvious enough but easy to lose sight of, is that these data represent

tensions only. A change in level of CO_2 curve would change the CO_2 tension also, providing pH_a values were constant.

6. When one considers the equilibrium levels in any two experiments done on the same morning or afternoon, the variations are smaller. There are five such groups, done on December 10, 14, 19, 21, and 31. In the cases of the first two of these, the subject was not at rest between experiments but got up and performed ordinary laboratory activity. The equilibrium (15 second) levels were 53.6 mm. and 52.9 mm. on December 10, and 52.8 mm. and 54.3 mm. on December 14. In the remaining three groups the subject remained at rest between experiments. The equilibrium levels were: 53.4 mm. and 52.6 mm. on December 19, 52.8 mm. and 53.4 mm. on December 21, and 54.8 mm. and 54.1 mm. on December 31. Thus it can be said that by this technique the apparent equilibrium values for successive experiments done within an hour, and with the subject at rest, differ by less than 1 mm.

7. Reference to the "Estimated mixed CO_2 tension" points in figure 4 will show that the equilibrium obtainable in the three groups II, III and IV corresponds in general with the fact that there is relatively little change necessary in the lung gases, after mixture, during the equilibration process. This is not so true of the 65 to 70 mm. group (IV) as the others, there being an excess of CO_2 in the lung air in these experiments. It is probable that lung tissue fluids are able to absorb appreciable amounts of this excess, in their change from alveolar to mixed venous tensions.

8. In the group in which the initial tensions in the bag were 70 mm. or over, there are also excellent plateaux formed, even flatter than in the preceding three groups, but their levels are definitely higher than the latter. Evidently the blood is unable to absorb all the excess of CO_2 in the lung-bag air before recirculation begins. It is to be noted that the plateau level rises only gradually above that of the lower groups, when the initial bag tension is increased from 70 to 80 mm. This has been explained in the theoretical discussion preceding.

9. There seems to be also a tendency for the CO_2 tension to rise less rapidly in the period from 25 to 35 seconds, when the initial CO_2 tensions in the rebreathing bag are high (see fig. 4). If this is true, it is difficult to explain: one would expect the opposite. Is it possible

that the arterial blood highly charged with CO₂ can dilate the tissue capillaries and venules to such an extent as to delay appreciably the recirculation of this particular part of the total blood volume? It is certainly true that the rebreathing of these higher CO₂ mixtures is associated with definite flushing of the skin and sensation of warmth.

The ten experiments performed on subject M. L. S., listed in table 5, give generally similar results to those just described. The CO₂ tension level in this subject was considerably lower than in the other, an average of 38.2 mm. for alveolar air, and of 46.7 mm. for mixed venous level. In the experiments in which the initial CO₂ tensions in the bag were 45 mm. and 48 mm., these values were too low for the establishment of an equilibrium level. In the six experiments in which the initial concentration in the bag was near 50 mm., it will be noted that the equilibrium level was in three cases attained in the 15 to 20 second interval, and in two cases in the 10 to 15 second interval. In one case the 15 second point was slightly below either of the two adjacent points. Apparently in this subject, recirculation of blood may take place in appreciable amounts in less than twenty seconds; the experiments are too few to decide this point. The two experiments with initial tensions of 60 mm. give plateau values definitely higher than the 50 mm. group. Probably this initial tension is beyond the equilibrating limit for this subject. This tension (60 mm.) is about 13 mm. higher than the apparent oxygenated mixed venous level; subject D. W. R. reached equilibrium with initial rebreathing mixtures up to 17 mm. above this level. This may have been due to the larger lung volume and residual air in the latter subject.

The definite equilibrium levels demonstrated above have, of course, been found by most workers on this subject in the past. The method which we have used, in establishing the limits of composition of rebreathed air, with which this equilibrium can be obtained, adds, we believe, further evidence in confirmation of their results.

(b) Oxygenated mixed venous tensions, method of successive rebreathings

Three experiments by this method were performed with subject D. W. R., according to the technique described above, each of these experiments being followed, after a fifteen minute interval, by another experiment of the series just described (single rebreathing

period), for comparison. In two of these, the breath was held for 18 seconds each time, in the third (as described under Methods) the mixture was rebreathed slowly four times in the 18 seconds. This difference of technique made no significant difference in results.

A comparison of the results by the two methods is given in table 6. It will be seen that the apparent "mixed venous" level by the successive rebreathing method was considerably lower than by the other method, in all three experiments; the level in the former group being 2 mm. or more lower than that in any of the 46 experiments by the other method. The slight lowering of CO₂ values after the fourth successive rebreathing period was due presumably to a lowering of O₂ tension, the venous blood not becoming completely oxygenated. In

TABLE 6
Oxygenated mixed venous CO₂ tensions by different methods

Date	A. Single rebreathing period						B. Successive rebreathings					
	Initial bag CO ₂ tension	Seconds after beginning rebreathing					Initial bag CO ₂ tension	Rebreathing period				
		10	15	20	25	30		1st	2nd	3rd	4th	5th
		CO ₂ tension						CO ₂ tensions				
	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.	mm.	
January 28.....	61.2	53.6	53.8	54.6	54.6	55.5	50.6	50.6	49.9	50.9	50.1	49.6
February 1.....	60.2	52.6	52.4	52.7	53.6	54.0	64.7	51.9	49.4	49.4	49.3	49.2
February 4.....	57.0	51.2	52.2	52.5	53.1	53.6	58.7	52.0	50.7	50.3	50.3	49.7

the experiment of February 1, the O₂ tension in the fourth sample was 110.0 mm., in the fifth sample 89.6 mm.

It is concluded that *for this subject*, the method of successive rebreathing from the same bag without enriching the mixture with CO₂, gives equilibrium values below those obtained by the single rebreathing method.

One similar experiment with subject M. L. S. seemed to show the same thing, but these data were not enough to decide the point.

(c) *True mixed venous blood tensions*

In order that a comparison may be made between mixed venous CO₂ content of blood as derived from oxygenated tension methods, and mixed venous CO₂ content of blood as derived by true tension methods

(equilibrating both CO_2 and O_2 simultaneously), the CO_2 - O_2 relationships must be known. For blood itself, these relations can, of course, be demonstrated by constructing a nomogram. If lungs and pulmonary vein blood are in approximate equilibrium at every moment during the rebreathing procedure, then the course of equilibration, as shown by successive CO_2 and O_2 tensions, can be plotted on this nomogram. A comparison can then be made between CO_2 content as obtained by the true and by the oxygenated method.

Making provisionally this assumption of approximate lung-blood equilibrium, we constructed a CO_2 - O_2 nomogram of the blood of D. W. R., and a similar but less complete nomogram of the blood of M. L. S.

TABLE 7
CO₂ dissociation curves of reduced and oxygenated blood. Subject D. W. R.

Oxygenated		Reduced		
Total CO_2	CO_2 tension	Total CO_2	CO_2 tensions	Total O_2
<i>volumes per cent</i>	<i>mm.</i>	<i>volumes per cent</i>	<i>mm.</i>	<i>volumes per cent</i>
43.5	31.1	46.4	26.2	1.9
43.7	31.2	60.4	52.1	1.7
53.3	48.5	64.2	54.4	0.9
60.9	65.6			

To describe first the nomogram of subject D. W. R.:— CO_2 curves of reduced and of oxygenated blood were made. The technique of blood-gas analysis which we used has been described in a previous paper (33); in general it follows the method developed by Austin, Van Slyke, et al. (34). The Van Slyke-Neill (35) manometric apparatus, and the Boothby modification of the Haldane gas analysis apparatus were used for blood and gas analyses respectively. The blood was prevented from clotting by the use of small amounts of heparin, and dry NaF was added, to give a final concentration of about 0.1 per cent, as a preservative. It was thought at first that this would prevent the loss of CO_2 capacity during equilibration in the tonometers. Following a suggestion kindly made by Dr. J. H. Talbott, however, we investigated this assumption, and found it incorrect, the blood losing about 0.6 cc. of CO_2 per 100 cc. blood, per hour of equilibration at 37.5°C . We, therefore, corrected our CO_2 content figures accordingly. Three

points were determined on the reduced and four on the oxygenated CO_2 curves, and the curves drawn by the use of the straight line logarithmic relation of Peters, Eisenman, and Bulger (36). The data for the curves are given in table 7.

The tonometers employed for the reduced CO_2 curve determinations actually contained small amounts of O_2 . This was measured in both blood and gas phases, and a correction readily obtained by plotting the actual points on a CO_2 - O_2 diagram, and with the aid of the oxygenated CO_2 curve, extrapolating to the line of zero oxygen content. By

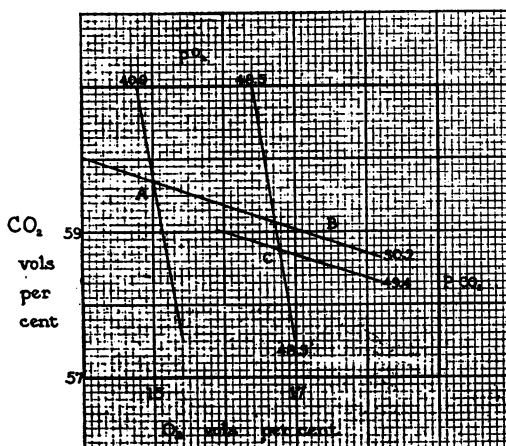


FIG. 5. EXPERIMENT OF MAY 13, 1929

Chart represents part of a CO_2 - O_2 nomogram of the blood of subject D. W. R. Point *B* = arterial blood at the end of 20 seconds rebreathing of mixed venous gas mixture. Point *C* = gas tensions in the lungs after 20 seconds' rebreathing.

the use of these extrapolated points, a completely reduced CO_2 curve was obtained.

In a previous study (33), complete oxygen dissociation curves on subject D. W. R. had been determined; these were used for the nomogram. Various workers have found that the O_2 dissociation curves of one individual change little from one year to the next (13, 37). One point was, however, determined again at an oxygen tension of 40 mm., and CO_2 tension of 50 mm.; this point, when plotted on the O_2 dissociation curve chart constructed three years previously, fell within 0.3

volumes per cent (of O₂) of its proper place; this is within the experimental error of the technique employed.

It may also be noted that the CO₂ curves which were made had prac-

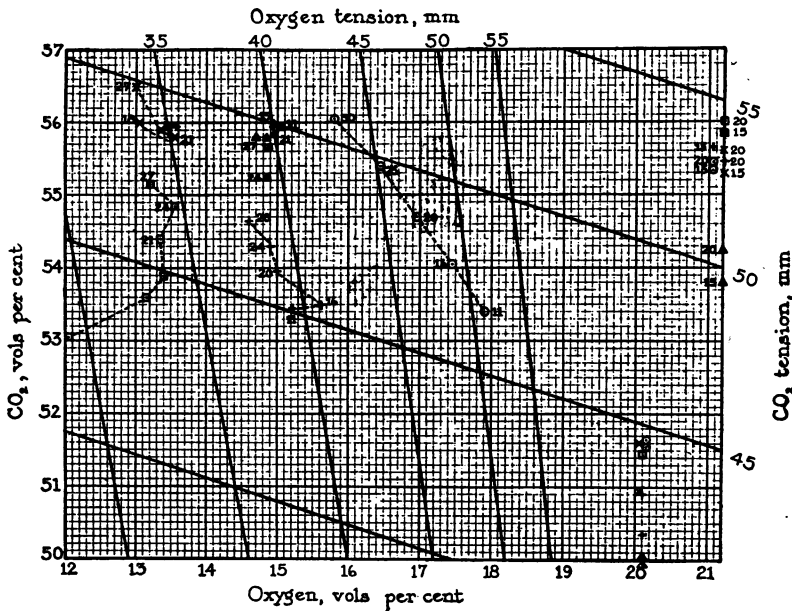


FIG. 6. CO₂-O₂ NOMOGRAM OF BLOOD OF SUBJECT D. W. R.

Plotted on it are the progressive changes, during the course of rebreathing, in CO₂ and O₂ tensions in lung-bag air, of five experiments. The figure beside each point represents the number of seconds after beginning rebreathing, at which that particular sample was taken. At the right of the diagram, along the completely oxygenated line, are the equilibrium (15 to 20 second) tensions of a corresponding series of oxygenated mixed venous experiments. Alveolar air tensions are plotted along the 95 per cent oxygen saturation (arterial) line. Each symbol represents one day's experiments, as follows: squares, February 13; crosses, February 16; triangles, February 18; plus marks, March 7; circles, March 11.

tically the same levels as had the CO₂ curves of this subject at the time the O₂ dissociation curves were studied.

The nomograms of the blood are used in figures 6, 7, and 8.

Before considering the rebreathing experiments, it is important to give the evidence which we have for the assumption of lung-blood

equilibrium during rebreathing. This experiment was performed on May 13, as follows:

The subject, after a half hour rest, applied mouthpiece and nose-clip and breathed through the apparatus (fig. 3). The two rebreathing bags had been filled with the following mixtures: Bag C, with 275 cc. CO₂ and 2,225 cc. N₂; Bag H with 225 cc. CO₂, 650 cc. of air, and 1,625

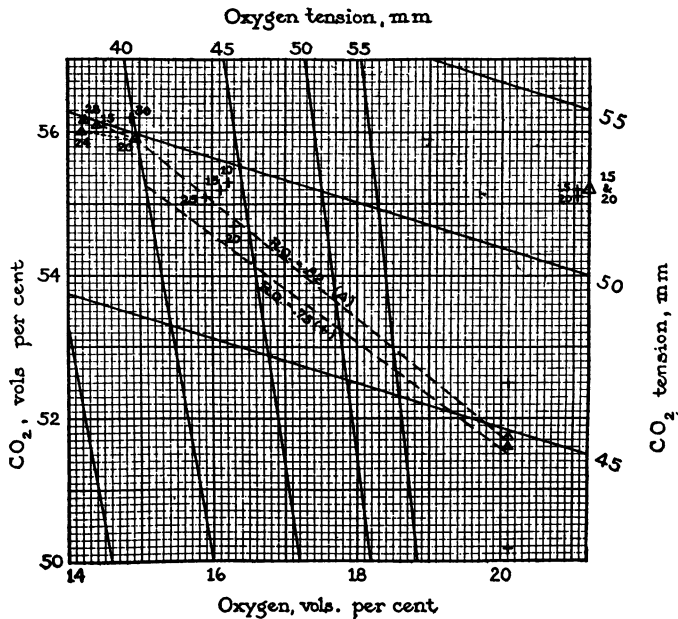


FIG. 7. CO₂-O₂ NOMOGRAM OF BLOOD OF SUBJECT D. W. R.

For explanation see figure 6, and text. Triangles = experiments of March 15. Plus marks = experiments of March 20.

cc. N₂. Just before the rebreathing procedure was begun, an arterial puncture was made into the brachial artery, but no blood drawn. The rebreathing was then performed, as described above. At the twentieth second of rebreathing, a signal was given and the assistant at the artery began drawing blood (under oil) and continued for the following twelve seconds. Rebreathing was continued by the subject during this interval, and the usual gas samples taken.

The blood was transferred under oil to a tube containing a small

amount of anticoagulant, dried neutral potassium oxalate being used in this instance (final concentration about 0.2 per cent), plus enough dried NaF to give a final concentration of 0.1 per cent. The arterial blood contents of CO₂ and O₂ were measured at once, and the remainder of the blood transferred to a tonometer containing about 50 mm. CO₂ and 40 mm. O₂. This blood was equilibrated at 37.5°C. for 50 minutes, and CO₂ and O₂ then determined on both blood and gas

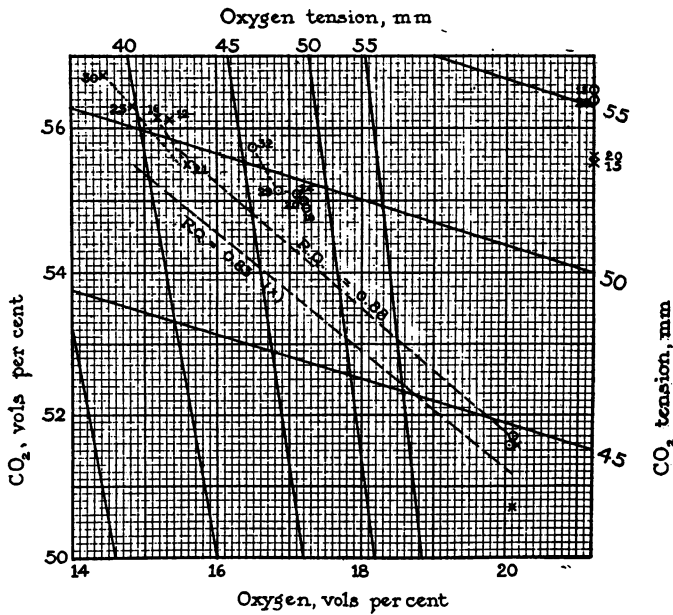


FIG. 8. CO₂-O₂ NOMOGRAM OF BLOOD OF SUBJECT D. W. R.

For explanation see figure 6, and text. Crosses = experiments of May 3. Circles = experiments of May 13.

phases. After correction had been made for CO₂ capacity lost during equilibration, the final figures were as shown in table 8. The tensions in the lung air during rebreathing were as shown in table 9.

It is evident that from the sixteenth to the twenty-fourth second, a nearly perfect equilibrium was apparently maintained in the lung gases. Inasmuch as it probably takes several seconds for the blood to go from lung alveoli to brachial artery, the arterial blood drawn from

the twentieth to the thirty-second second must have come near being the blood which passed through the lung alveoli during the interval of equilibrium.

If equilibrium were perfect between lungs and blood, and technique of blood handling, etc., were faultless, then the arterial blood gas contents, and the lung tensions, when plotted on a $\text{CO}_2\text{--O}_2$ nomogram, should fall on the same point. Such a nomogram is given in figure 5, point *A* being that of the blood and gases in the tonometer, and the tension lines of this diagram drawn with distances and slopes

TABLE 8
Evidence of lung-blood equilibrium during rebreathing

Arterial blood contents		Blood equilibrated at 37.5°C.			
CO ₂	O ₂	CO ₂		O ₂	
<i>volumes per cent</i>	<i>volumes per cent</i>	<i>volumes per cent</i>	<i>mm.</i>	<i>volumes per cent</i>	<i>mm.</i>
59.0	17.4	59.7	50.2	15.0	40.9

TABLE 9
Tensions in lung air, same experiment as table 8

Time after starting	CO ₂	O ₂
<i>seconds</i>	<i>mm.</i>	<i>mm.</i>
16	49.3	48.7
20	49.4	48.5
24	49.5	48.1
29	49.6	47.0
32	50.5	45.5

corresponding to those of the more complete nomogram of this subject's blood, previously described. This is probably a justifiable extrapolation, over so small an area of the nomogram. Point *B* is the "mixed venous" blood (as drawn from brachial artery), and *C* the lung tensions after 20 seconds' rebreathing. It will be seen that the differences between the two points (*B* and *C*) are small: 0.25 volumes per cent, or 0.8 mm. of CO_2 , and 0.6 volumes per cent, or about 3 mm. of O_2 . Considering the amount of manipulation of blood necessary, this agreement is fairly good. The chief discrepancy is in oxygen, the arterial blood value being too high. The slow diffusion rate of

oxygen as compared with that of CO_2 is probably an important factor in delaying this equilibrium.

We ought to have done more experiments of this sort. It was not easy, however, to time the procedure accurately, and several unsuccessful attempts were made.

It may be concluded from this experiment that when the lung gases are at equilibrium during rebreathing (in the region of mixed venous levels), there is also approximate equilibrium between these gases and the blood leaving the pulmonary capillaries.

One cannot, of course, conclude that there is equilibrium at any moment between lungs and pulmonary vein blood, when the pulmonary artery blood varies considerably from lung tensions. However, for small lung-blood differences the discrepancy will be relatively small. One can be justified, therefore, in using blood nomograms to represent the course of equilibration during lung-blood experiments. Their direction of change will be properly represented, and if an apparent equilibrium level is reached (as indicated by constant lung tensions) then it will be known that lungs and blood have nearly the same tensions.

In the case of oxygenated mixed venous tension experiments, the demonstration of equilibrium between lungs and blood after 15 seconds' rebreathing was made, as noted above, by Field, Bock, et al. (16).

Figures 6, 7, and 8 are CO_2 - O_2 nomograms of the blood of D. W. R. on which are plotted the CO_2 and O_2 tensions of samples of rebreathed mixtures taken at successive intervals after the beginning of rebreathing using the technique described under Method (c). Each experiment is represented by a series of four to six points, connected by a dotted line, the number beside each point indicating the number of seconds after starting rebreathing, at which the sample was taken. At the extreme right, along the line of complete oxygen saturation are given, for purposes of comparison, the levels, in CO_2 tensions, of the fifteenth and twentieth second samples of the oxygenated mixed venous tension experiments. As stated above, one of these was done, in every instance, after the true mixed venous tension experiment had been completed. There may be a small error involved here, as the blood hemoglobin even with the rebreathing of high oxygen mixtures, may not become completely saturated (see Part I). Along the 95 per cent

saturation line are plotted the CO_2 tensions of alveolar air samples (corresponding to arterial blood CO_2 tensions).

It is clear that if both methods ("true" and "oxygenated") actually give at equilibrium the tensions of the incoming blood, and the latter remain unchanged during the procedures, then both equilibrium points on the nomogram will lie on the same CO_2 content line.

Each pair of experiments done on the same day (one "true" and one "oxygenated") are presented on the figures by one symbol (triangle, cross, etc.).

Perfect equilibrium on the nomogram is, of course, represented by the exact superposition of two or more successive samples.

On figure 6 are given the results of five experiments, in four of which, so far as the true mixed venous tensions were concerned, equilibrium was evidently not arrived at.

In the experiment of February 13, for example, the initial mixture in the rebreathing bag was too low both in CO_2 and O_2 . The CO_2 progressively increased during rebreathing, and the O_2 at first increased and then, after the twenty-fourth second, diminished, indicating recirculating blood. The oxygenated tension experiment reached a plateau value at 54.0 mm., as shown. This corresponds to 55.8 volumes per cent of CO_2 , well above any of the points on the other curve.

Similarly, the experiment of March 11 shows that the initial mixture was too high in O_2 and too low in CO_2 . During rebreathing the CO_2 tension progressively increased and O_2 tension diminished. The plateau level of CO_2 content in the oxygenated tension experiment was 55.4 volumes per cent. This value for CO_2 was also reached in the "true" mixed venous experiment, but only after twenty-five seconds, a time when recirculation of blood had probably begun.

The experiment of March 7 is similar to that of March 11.

In the experiment of February 16, the three successive samples taken at 18, 21, and 24 seconds have the same CO_2 tensions and thus indicate an equilibrium with respect to this gas. Oxygen tensions showed an increase up to 21 seconds, followed by a decrease, suggestive evidence that the oxygen tension at 21 seconds was still below the equilibrium value. Further evidence in favor of this may be obtained by drawing an average R.Q. line (0.81) on the nomogram, starting from the alveolar air tension (as plotted on the 95 per cent O_2

saturation line). This line will be found to fall well to the right of the 18, 21 and 24 second points.

It will be noted that the 18 and 21 second CO_2 levels were higher by about 0.4 volume per cent than the plateau values in the corresponding oxygenated tension experiment. This is probably due in part at least to the failure to equilibrate with respect to oxygen.

On February 18, apparently an equilibrium was reached and maintained several seconds, as shown by the close grouping of the points. Unfortunately it is not possible to compare the equilibrium value for the CO_2 level with that of the corresponding oxygenated tension experiment, because the initial mixture used in the bag in the latter contained only 52.5 mm. CO_2 , and a plateau level was, therefore, not reached in the fifteenth to the twentieth second. The level of the true mixed venous CO_2 is, however, close to, though slightly above, the average level of equilibrium arrived at in other oxygenated tension experiments. There is a further difficulty with this experiment and that is the unusually low alveolar air tensions. These, when compared with the equilibrium point for the true mixed venous tension, give an impossibly high value for the R.Q. Where the error, or errors, occurred in this experiment we have been unable to discover.

In figures 7 and 8 are given four experiments in which a fairly close approach to equilibrium was achieved, as indicated both by the close grouping of the points, and their nearness to the R.Q. lines (slopes as determined by analyses of expired air, lines drawn from alveolar air points on figure). It will be seen that the points are clustered around the same CO_2 content lines as are the corresponding equilibrium values for the oxygenated tension experiments (plotted at the right of the nomogram). On May 13, the initial tension in the bag for the oxygenated tension experiment was over 70 mm., so that the tension at the "plateau" level, 55.3 mm., was almost certainly too high. Disregarding this experiment, the differences in CO_2 content level between the true and the oxygenated equilibrium values (average of fifteenth and twentieth second samples) were as follows: March 15, +0.7 volumes per cent (i.e., "true" higher than "oxygenated"); March 20, -0.2 volumes per cent; May 3, +0.3 volumes per cent, an average of +0.3 volumes per cent.

There seems to be a tendency for the true equilibrium values to

fall somewhat to the right of their respective R.Q. lines. The reason for this is not clear.

It will be noted in these true mixed venous experiments, that equilibrium is much more readily obtained with respect to CO_2 than O_2 . This is to be expected, and has been considered in the theoretical discussion preceding. Both the small amounts of oxygen available from the blood, and the slow diffusion rate of this gas, will delay equilibrium.

We may note also that the nine experiments above described are all that were done with this technique on this subject.

We had hoped to find some evidence as to whether the concentrating effect of oxygen absorption (see Part I) would make the equilibrium CO_2 content levels of the oxygenated tension experiments higher than those of the corresponding true mixed venous tension group. The evidence, such as it is, is against this, but our data were too inexact and disperse to admit of a proper answer to this question. It is quite possible that the blood in the pulmonary capillaries after rebreathing the $\text{CO}_2 - \text{O}_2$ mixtures, was not more than 97 or 98 per cent saturated with oxygen, instead of 100 per cent, in which case this would tend to offset the error due to oxygen absorption.

It may, however, fairly be concluded from this group of experiments that:

1. Equilibrium of lung-bag tensions with the tensions of the incoming venous blood can be achieved with respect to CO_2 and O_2 simultaneously, only if the mixed lung gases are, at the time when rebreathing is begun, within a very few millimeters of the actual mixed venous tensions. In other words, little adjustment can be made by the absorption or liberation of gases by the blood.

2. When true mixed venous equilibrium has been attained, these values, in respect to blood CO_2 contents are at the same general level, and in each particular instance close to the corresponding value arrived at by a properly conducted oxygenated mixed venous tension determination. By "properly conducted," in the case of this subject it is meant that the initial tension of CO_2 in the rebreathing bag shall be between 56 and 70 mm.

The correspondence between the mixed venous CO_2 contents as arrived at by these different methods, coupled with the definiteness with which an apparent equilibrium can be reached by both methods,

strengthens materially, we believe, the inference that the CO₂ tensions so obtained are actually the CO₂ tensions of the mixed venous or pulmonary artery blood, at the moment when rebreathing is completed. The experiments confirm the conclusions reached in 1922 by Douglas and Haldane (13), but have extended their observations somewhat, in showing the process of equilibration more fully, and in giving more definite justification for the assumption of lung-blood equilibrium for respiratory gases.

It is of interest to calculate the volume output of the heart of this subject on the basis of our data. From analyses of expired air, on February 18, March 7, 11, 15, and May 3, the corrected figures for CO₂ output were 217.5 cc., 198.3 cc., 223.8 cc., 222.7 cc., and 204.5 cc., an average of 213.4 cc. From table 7, the average alveolar CO₂ tension is found to be 43.9 mm., and the average oxygenated mixed venous CO₂ tension of groups II, III, and IV of this table, at the end of fifteen seconds' rebreathing, is 53.3 mm. On the CO₂ dissociation curve of this subject, 43.9 mm. corresponds to 51.3 volumes per cent, and 53.3 mm. to 55.5 volumes per cent. The volume output of the heart is thus

thus $\frac{213.4}{(55.5 - 51.3) \times 10} = 5.1$ liters per minute. As this subject has a surface area of 2.00 square meters, the cardiac output per square meter is 2.6 liters. This is higher than 2.2 liters, the normal standard which Grollman (28) has worked out by the use of his acetylene method of measuring circulation rate; considering the fact that conditions in our experiments were not rigidly basal, the agreement is fairly close.

The individual figures for circulation rate, as derived from the "true" mixed venous blood experiments, show considerable variation, due chiefly to variations in alveolar air values. The mixed venous CO₂ content values for the five experiments in which an apparent equilibrium was reached were 55.9, 55.9, 55.7, 55.0, and 54.9 volumes per cent, an average of 55.5 volumes per cent,—the same as for the oxygenated experiments. The corresponding alveolar air samples (41.6, 44.4, 46.3, 44.5, and 44.7 mm.) gave an average arterial CO₂ content of 51.5 volumes per cent. The circulation rate is then $\frac{213.4}{(55.5 - 51.5) \times 10} = 5.3$ liters per minute.

In the case of subject M. L. S., two experiments were performed, each consisting of one true and one oxygenated equilibration. In order that the results might be compared, a nomogram was constructed, as described above. In this instance, however, the number of points actually determined was small: two points on the oxygenated CO_2 curve, one on the reduced curve, one point at 40 mm. CO_2 and 30 mm.

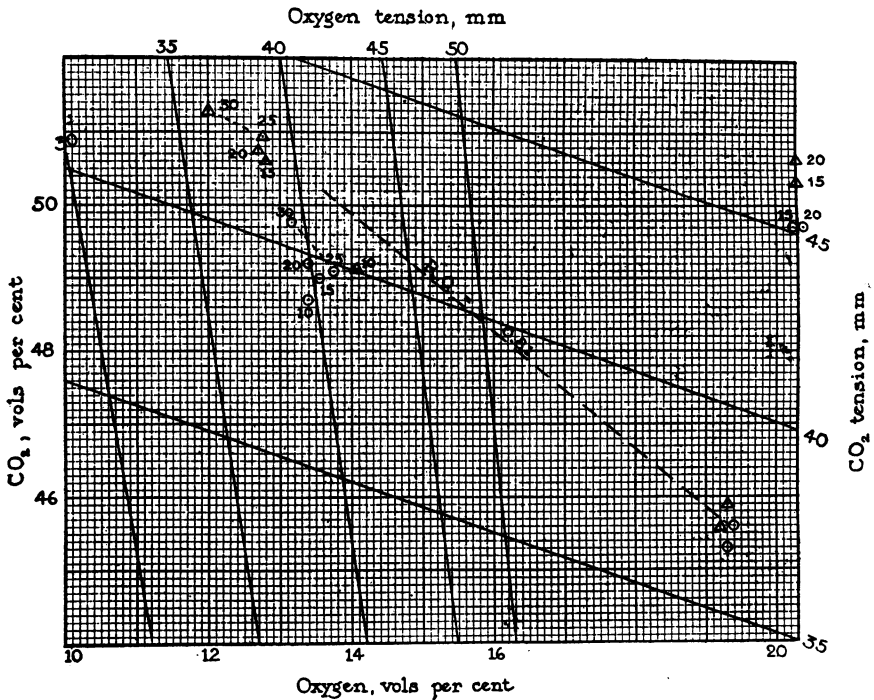


FIG. 9. CO_2 - O_2 NOMOGRAM OF BLOOD OF SUBJECT M. L. S.

For explanation see figure 6, and text. Circles = experiments of February 25. Triangles = experiments of March 1.

O_2 tensions, and one at 40 mm. CO_2 and 45 mm. O_2 tensions. These were sufficient to give the skeleton of a complete nomogram. The CO_2 tension lines were drawn by using slopes and distances similar to those of the nomogram of D. W. R.; the error involved by this procedure, over the small area under consideration, is probably not great.

The nomogram is given in figure 9, with the two pairs of experiments

plotted on it. Expired air samples were not obtained, but average R.Q. lines ($\bar{V} = 0.8$) have been drawn from each of the alveolar air points. It will be seen that the points in each true venous tension experiment are fairly close together. The agreement in CO_2 content level between the true and corresponding oxygenated points at equilibrium is fair: the average of fifteenth and twentieth second points show a difference of -0.6 volumes per cent ("true" less than oxygenated) in the first experiment, and $+0.25$ volumes per cent in the second. From the position of the R.Q. lines, the latter would appear to be closer to the true mixed venous point.

As in the experiments with subject D. W. R., it can be said that "true" and oxygenated methods appear to give the same general level of venous CO_2 content. The average oxygen utilization or arteriovenous difference, for this subject was 5.6 volumes per cent.

The recent experiments of Lauter with Bauman and Friedländer (25), in which blood was taken directly from the right hearts of human subjects in the resting basal state, showed the average blood flow to be about 4.5 liters per minute, and the average arteriovenous oxygen difference 5 volumes per cent. This compares well with the above results on the two subjects of our experiments, under non-basal conditions. As we studied only two subjects, no conclusion can be drawn as to the absolute accuracy of the results we have obtained, but this comparison of indirect with direct mixed venous blood determinations suggests that the unproved assumptions (see Part I) which we have made in these experiments—particularly relative to respiratory movement and recirculation of blood—do not involve any large error.

SUMMARY

From the data presented in this paper, the following summary of results obtained may be given:

1. For the investigation of mixed venous blood gases in man at rest, two normal subjects were studied. The methods employed were simple, involving the rebreathing of gas mixtures from a bag, and the taking of instantaneous samples at intervals of 5 seconds during the rebreathing period. The latter lasted usually about 30 seconds, sometimes as long as 50 seconds.

2. In the experiments in which oxygenated mixed venous blood was studied, it was found that a "plateau" level for CO_2 tensions could be obtained regularly near the fifteenth second after beginning rebreathing, providing suitable initial CO_2 tensions were used in the rebreathing bag. These tensions could be varied within well defined limits, the lower limit being a value about 2 mm. higher than the resulting equilibrium level; and the upper limit about 15 mm. above this level in the case of one subject, and 10 mm. above in the case of the other. The equilibrium level was on the average practically the same for all bag mixtures used, within these limits.

3. Below this lower limit of initial CO_2 tension, no constant plateau level was reached during rebreathing. Above the upper limit of initial CO_2 tension, plateaux were found which were definitely above those reached by the lower tensions.

4. The plateaux, or equilibrium levels of CO_2 tension, were not quite horizontal, but showed slight upward inclination. It was thought that this might be due in part to progressive oxygen absorption during the interval, and in part to recirculation of small volumes of blood.

5. The definiteness of the plateau or equilibrium levels was indicated by the facts (*a*) that they occurred regularly over the same time interval, (15 to 20 seconds), (*b*) that the deviations in slope of individual experiments from the average were much less over this interval than over earlier and later intervals, and (*c*) that widely different initial concentrations of CO_2 in the rebreathing bag were brought to the same equilibrium level at the end of rebreathing.

6. The actual equilibrium levels of mixed venous CO_2 varied with the time of day, and varied from one day to another. Successive experiments on the same morning with the subject remaining at rest, gave levels varying by less than 1 mm. of CO_2 .

7. In the experiments in which it was attempted to establish true mixed venous equilibrium (in respect to both CO_2 and O_2 tensions) between lungs and blood, it was found that the mixed lung and bag gases had to be close to the actual venous tensions, in order for equilibrium to be established in the lung gases before recirculation of blood began.

8. One experiment in which arterial blood was drawn while such a rebreathing experiment was being done, showed that, near the mixed

venous level, at the time when the lung gases were in apparent equilibrium, lung tensions and blood tensions were nearly the same.

9. By plotting on a blood nomogram oxygenated and true mixed venous tension experiments, it was found that successive experiments done by the two methods gave approximately the same venous CO₂ contents at the equilibrium levels.

10. The time when appreciable amounts of blood began to recirculate, in the cases of the two subjects of these experiments, was usually between 20 and 25 seconds. This was indicated by (a) the time when oxygenated CO₂ tensions began to show a sharp upward trend, after the plateau level; (b) the time when oxygen tensions, in the true mixed venous tension experiments, after an initial steady increase, began to decrease again.

11. Experiments in which successive rebreathing periods were employed, leaving the bag mixture unchanged between periods, gave equilibrium levels definitely lower than either of the other two groups of experiments, in the case of one of the two subjects. It was concluded that *for this subject*, the successive rebreathing method did not give mixed venous values. It was suggested also that this might account for some of the high values for circulation rate obtained with circulation rate methods in which this technique was used.

12. Calculations of circulation rate, in one of the two subjects, gave the following average values: (a) from the oxygenated mixed venous CO₂ tension values, 5.1 liters per minute; (b) from the true mixed venous CO₂ values, 5.3 liters per minute. Circulation rate per square meter of body surface (conditions not basal), 2.6 liters per minute.

13. The arteriovenous oxygen difference for one subject was 5.1 volumes per cent, for the other 5.6 volumes per cent. This compares closely with the values obtained by Lauter, by direct cardiac puncture in man.

CONCLUSION

1. By the use of a simple technique of rebreathing CO₂-O₂-N₂ mixtures, with samples taken every 5 seconds during the procedure, the course of equilibration of lung air with incoming venous blood has been studied. It has been found that an equilibrium can be regularly

established, for CO₂ with certain high oxygen mixtures, and for both CO₂ and O₂ with certain low oxygen mixtures. These two equilibria represent essentially the same CO₂ content levels in the (venous) blood of the subject. The mixtures of rebreathed gases that can be used to establish such equilibria can be varied within small but fairly well defined limits.

2. To accept these equilibria as representing the state of the mixed venous blood at rest involves certain assumptions; these have been described in the first part of the present study. Previous workers have shown that (*a*) with proper technique complete mixture of lung-bag gases is attainable; (*b*) "true" and "oxygenated" mixed venous CO₂ levels are the same. Our experiments have confirmed the above and have further shown that (*c*) the error due to progressive oxygen absorption during the oxygenated mixed venous equilibration is small, (*d*) lung gases and blood are in approximate equilibrium at the end of a "true" mixed venous equilibration, (*e*) the method of intermittent rebreathings gives false "mixed venous" equilibria in one of the subjects studied.

3. No further evidence has been obtained as to the presence or extent of error due to (*a*) altered respiratory movements during the procedure, or (*b*) recirculation within the time of the procedure, of small amounts of blood; but the comparison of our arteriovenous differences with those recently obtained by Lauter with direct cardiac puncture, suggests that the error involved here is small.

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